

ANSWER KEY

ATOMS AND ELECTRONS STUDY GUIDE

Learning Goals: (a) (a.i) P. 4.1 - Explain a new scientific theory of atomic scale phenomena (atomic theory) by describing changes in the atomic model over time, including those changes that are consistent with experimental evidence. (a.i.ii) P. 4.2 - Explain the scientific theory of atomic scale phenomena, including the scientific theories in terms of particles, motions, interactions, and the various forces that act to hold a system of these particles together, and describe, within the atom, (a.i.iii) P. 4.3 - Explain properties of atoms and their motions in the various media, for example, in terms of their interactions, related to P. 4.4, in terms of the scientific evidence of their first fundamental (scientific) properties, (a.i.iii) P. 4.4, in terms of their properties, such as mass, size, and shape.

ANSWER KEY

Directions: Read the study theory selected on the correct reading.

ANSWER	ANSWER KEY	ANSWER KEY: STUDY THEORY
1C	Correctness (a.i.ii) P. 4.2	1. Everything is made of atoms that are in motion, moving, and interacting.
2A	Correct (a.i.ii) P. 4.2	2. These findings show that atoms are in motion, moving, and interacting with each other through the various forces.
3B	Correct (a.i.ii) P. 4.2	3. Atoms are in motion (moving).
4D	Correct (a.i.ii) P. 4.2	4. Atoms are in motion (moving).
5C	Correct (a.i.ii) P. 4.2	5. Atoms are in motion (moving).
6B	Correct (a.i.ii) P. 4.2	6. Atoms are in motion (moving).

Directions: Use the provided evidence to answer the following questions:

- 1C** - P. 4.2, 4.3 - The number of particles equals the number of electrons.
- 2A** - P. 4.2, 4.3 - The number of particles equals the number of electrons.
- 3B** - P. 4.2, 4.3 - The number of particles equals the number of electrons.

Study Guide Why Atoms Combine Answer Key

LL Leslie

A decorative graphic consisting of a light blue horizontal bar with a rounded right end, and a red circular shape partially visible behind it.

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