

2. Do you think a chemical reaction has taken place? Explain.

Yes, I do think that the gas discharged into the air caused a chemical reaction to occur.

**Activity A:**

**Observing  
chemical changes**

Get the Gizmo ready:

- Click **Reset** (↺). Check that the reactants are still **Sodium** and **Water**.
- Turn on the **Label reactants** checkbox.



**Introduction:** It is important to distinguish chemical changes, in which new substances are formed, from **physical changes**, which do not create new substances. In this activity, you will look at many kinds of evidence that chemists use to see if a chemical change has taken place.

**Question:** What kinds of evidence indicate a chemical change has taken place?

1. **Observe:** Some chemical reactions release heat, and others absorb heat. In an **exothermic** reaction, heat is released and the temperature of the system rises. In an **endothermic** reaction, heat is absorbed and the temperature of the system decreases.

In the Gizmo, drag the **Thermometer** into the flask of water.

- A. What is the starting temperature?
- B. Click **Play**, and wait for the reaction to end. What is the final temperature?
- C. Was this reaction exothermic or endothermic?

21 Celsius

59 Celsius

Exothermic.

2. **Observe:** Two families of chemicals are **acids** and **bases**. Acids and bases can be detected by an **indicator**, which is a substance that changes color in the presence of an acid or a base. Phenol red is an indicator that is yellow in an acid, orange in a neutral solution, and pink in a base.

- A. Click **Reset**. Drag the **Phenol red** next to the flask of water. What does the indicator show?\_\_

Start PH: Neutral, 423.0g mass. Orange Phenol.

- B. Click **Play**, and wait for the reaction to end. What does the indicator show now?

PH: Basic, 422.9g mass, sodium dissolved, When the water from the sodium was added, phenol turned red or pink.

3. **Observe:** Click **Reset**. Select the **Gas collection** setup. Chemists use this apparatus to collect any gases produced in the reaction. From the reaction flask, gases travel through a long tube and into a cylinder of water. As gases bubble into the cylinder, the water is displaced (removed) until the cylinder is filled with gas.

Click **Play** and observe the cylinder. Was any gas produced in the reaction?

# Study Of Chemical Changes Lab Answer Key

**Ensheng Dong**



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### **Table of Contents Study Of Chemical Changes Lab Answer Key**

1. Understanding the eBook Study Of Chemical Changes Lab Answer Key
  - The Rise of Digital Reading Study Of Chemical Changes Lab Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Study Of Chemical Changes Lab Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Study Of Chemical Changes Lab Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Study Of Chemical Changes Lab Answer Key

- Personalized Recommendations
  - Study Of Chemical Changes Lab Answer Key User Reviews and Ratings
  - Study Of Chemical Changes Lab Answer Key and Bestseller Lists
5. Accessing Study Of Chemical Changes Lab Answer Key Free and Paid eBooks
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    - Study Of Chemical Changes Lab Answer Key Compatibility with Devices
    - Study Of Chemical Changes Lab Answer Key Enhanced eBook Features
  7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Study Of Chemical Changes Lab Answer Key
    - Highlighting and Note-Taking Study Of Chemical Changes Lab Answer Key
    - Interactive Elements Study Of Chemical Changes Lab Answer Key
  8. Staying Engaged with Study Of Chemical Changes Lab Answer Key
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Study Of Chemical Changes Lab Answer Key
  9. Balancing eBooks and Physical Books Study Of Chemical Changes Lab Answer Key
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Study Of Chemical Changes Lab Answer Key
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Study Of Chemical Changes Lab Answer Key
    - Setting Reading Goals Study Of Chemical Changes Lab Answer Key
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Study Of Chemical Changes Lab Answer Key

- Fact-Checking eBook Content of Study Of Chemical Changes Lab Answer Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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