

Chapter 21: Nuclear Chemistry

1. What is a nuclide?
2. What subatomic particle identifies a nuclide?
3. In carbon-12, what does the 12 represent?
4. Define mass defect.
5. What does mass defect represent?
6. What is Einstein's famous equation?
7. What kinds of elements have the highest nuclear binding energies?
8. How does the stability of a nuclide relate to the nuclear binding energy?
9. On the band of stability, what is the neutron-proton ratio for low atomic number atoms?
10. How does the neutron-proton ratio change as the atomic number increases?
11. Unstable nuclides undergo what type of change?
12. **True or False:** As a radioactive atom undergoes radioactive decay, the atom gives off large amounts of energy and increases its stability.
13. Define nuclear fission.
14. Define nuclear fusion.
15. During radioactive decay, the nucleus disintegrates into a (lighter or heavier) nucleus that is (less stable or more stable)
16. How does alpha decay affect the nucleus?
17. How does beta decay affect the nucleus?
18. How does gamma radiation affect the nucleus?
19. How does positron emission affect the nucleus?
20. Which form of radiation has the greatest penetration power?
21. Put the following in order from greatest mass to least mass: beta, alpha, gamma.
22. Alpha particles are (a) electrons (b)electromagnetic waves (c) helium nuclei
23. Beta particles are (a) electrons (b)electromagnetic waves (c) helium nuclei
24. What is the symbol for an alpha particle?
25. What is the symbol for a beta particle?
26. What is the symbol for a positron?
27. What is the symbol for a gamma ray?
28. Define half-life.
29. What can we say about a nuclide with a short half-life compared to one with a long half-life?
30. Define decay series.
31. What is the difference between natural transmutation and artificial transmutation?

Solve the following problems:

1. How much of a 35.5 g sample of radon-222 is leftover after 30.4 days? The half-life of radon-222 is 3.8 days.

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CH Cherryholmes



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Table of Contents Topic 12 Nuclear Chemistry Review Questions Answers

1. Understanding the eBook Topic 12 Nuclear Chemistry Review Questions Answers
 - The Rise of Digital Reading Topic 12 Nuclear Chemistry Review Questions Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Topic 12 Nuclear Chemistry Review Questions Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Topic 12 Nuclear Chemistry Review Questions Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Topic 12 Nuclear Chemistry Review Questions Answers
 - Personalized Recommendations
 - Topic 12 Nuclear Chemistry Review Questions Answers User Reviews and Ratings

- Topic 12 Nuclear Chemistry Review Questions Answers and Bestseller Lists
- 5. Accessing Topic 12 Nuclear Chemistry Review Questions Answers Free and Paid eBooks
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 - Topic 12 Nuclear Chemistry Review Questions Answers Budget-Friendly Options
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 - ePub, PDF, MOBI, and More
 - Topic 12 Nuclear Chemistry Review Questions Answers Compatibility with Devices
 - Topic 12 Nuclear Chemistry Review Questions Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Topic 12 Nuclear Chemistry Review Questions Answers
 - Highlighting and Note-Taking Topic 12 Nuclear Chemistry Review Questions Answers
 - Interactive Elements Topic 12 Nuclear Chemistry Review Questions Answers
- 8. Staying Engaged with Topic 12 Nuclear Chemistry Review Questions Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Topic 12 Nuclear Chemistry Review Questions Answers
- 9. Balancing eBooks and Physical Books Topic 12 Nuclear Chemistry Review Questions Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Topic 12 Nuclear Chemistry Review Questions Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Topic 12 Nuclear Chemistry Review Questions Answers
 - Setting Reading Goals Topic 12 Nuclear Chemistry Review Questions Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Topic 12 Nuclear Chemistry Review Questions Answers
 - Fact-Checking eBook Content of Topic 12 Nuclear Chemistry Review Questions Answers
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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