

NAME _____ INDEX NUMBER _____
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SALTS

1. 1989 Q 34b,c

- (a) Explain how sodium carbonate can be used to soften hard water.
(Use ionic equations where necessary). (3 marks)

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- (b) Give one other commercial use of sodium carbonate, besides softening of hard water. (1 mark)

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2. 1989 PPIA QUESTION 7

What would be observed when aqueous sodium hydroxide is added to aqueous lead (II) nitrate?

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3. 1989 PPIA QUESTION 18

P, Q and R are solutions of lead nitrate, sodium chloride and magnesium Sulphate (not in that order) Use the following information to identify P and Q

- (i) When drops of barium chloride solutions are added to separate portions of P, Q and R, a white P is formed
- (ii) When dilute hydrochloric acid is added to separate portions of Q and R, a white precipitate is observed in R but Q remains clear. (1 mark)

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