



Toyota Prius Uk Navigation Manual

James Kalbach



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 $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$
 $2\text{NaCl} + \text{F}_2 \rightarrow 2\text{NaF}$
 $\text{Cl}_2 + 4\text{H}_2\text{O} \rightarrow 2\text{H}_2 + 2\text{H}_2\text{O}_2$
 $5\text{Pb} + 2\text{HCl} \rightarrow 2\text{H}_2 + \text{PbCl}_2$
 $6\text{AlBr}_3 + 3\text{K}_2\text{SO}_4 \rightarrow 4\text{KBr} + \text{Al}_2\text{SO}_4 + 3\text{Br}_2$
 $4\text{CH}_4 + 2\text{O}_2 \rightarrow 2\text{CO}_2 + 2\text{H}_2\text{O}$
 $8\text{C}_3\text{H}_8 + 5\text{O}_2 \rightarrow 3\text{CO}_2 + 4\text{H}_2\text{O}$

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 $1\text{O}_2 + 3\text{H}_2 \rightarrow 3\text{H}_2\text{O}$
 $2\text{O}_2 + 2\text{H}_2\text{O} \rightarrow 2\text{H}_2 + \text{O}_2$
 $4\text{RbH} + 2\text{O} \rightarrow \text{RbOH} + 2\text{H}_2$
 $5\text{Cl}_2 + 2\text{KBr} \rightarrow \text{KCl} + \text{Br}_2$
 $6\text{N}_2 + 2\text{H}_2 \rightarrow 3\text{N}_2 + 7\text{H}_2$
 $3\text{Ca} + \text{N}_2 \rightarrow \text{Ca}_3\text{N}_2$
 $2\text{CH}_4 + \text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$
 $9\text{C}_2\text{H}_5\text{OH}$

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 $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$
 $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$
 $2\text{NaCl} + \text{F}_2 \rightarrow 2\text{NaF}$
 $\text{Cl}_2 + 2\text{H}_2\text{O} \rightarrow 2\text{H}_2 + \text{O}_2$
 $2\text{H}_2\text{O} + \text{Pb} \rightarrow \text{PbOH}_2 + 2\text{HCl}$
 $2\text{H}_2\text{O} + \text{PbCl}_2 \rightarrow \text{AlBr}_3 + 3\text{K}_2\text{SO}_4 \rightarrow 6\text{KBr} + \text{Al}_2\text{SO}_4 + 3\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$

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 $4\text{Fe} + 3\text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3$
 $2\text{FeBr}_3 + 3\text{H}_2\text{SO}_4 \rightarrow 1\text{Fe}_2(\text{SO}_4)_3 + 3\text{H}_2\text{O} + 3\text{H}_2\text{SO}_4$
 $1\text{C}_4\text{H}_{10} + 6\text{O}_2 \rightarrow 4\text{CO}_2 + 5\text{H}_2\text{O}$

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 $38\text{Fe} + 2\text{O}_2 \rightarrow 3\text{Fe}_2\text{O}_3$
 $2\text{Fe} + 3\text{CO} \rightarrow 2\text{Fe}_2\text{O}_3 + 3\text{CO}_2$
 $39\text{H}_2 + 3\text{PO}_4 \rightarrow 3\text{BaO} + 3\text{PO}_4$
 $4\text{H}_2 + 2\text{O}_2 \rightarrow 4\text{H}_2\text{O}$
 $40\text{Mg} + \text{NH}_4\text{PO}_4 \rightarrow 4\text{Mg}_2\text{P}_2\text{O}_7 + 2\text{NH}_3 + 2\text{H}_2\text{O}$
 $41\text{Al} + 3\text{H}_2\text{SO}_4 \rightarrow 2\text{Al}_2\text{O}_3 + 3\text{H}_2 + 2\text{SO}_4$
 $42\text{P}_4 + 10\text{H}_2\text{O} \rightarrow 3\text{H}_3\text{PO}_4 + 4\text{H}_2$
 $43\text{H}_2 + 2\text{O}_2 \rightarrow 3\text{H}_2\text{O} + 2\text{H}_2\text{O}$

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