

Section 1: Compounds (Library: Muller (Muller))

Trisubstituted benzene (C₆H₃)

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$$C_6H_3 = C_6H_5 - C_6H_4 - C_6H_3 - C_6H_2 - C_6H - C_6$$

1. The general formula for trisubstituted benzene is C₆H₃ = C₆H₅ - C₆H₄ - C₆H₃ - C₆H₂ - C₆H - C₆

$$C_6H_3 = C_6H_5 - C_6H_4 - C_6H_3 - C_6H_2 - C_6H - C_6$$

2. The general formula for trisubstituted benzene is C₆H₃ = C₆H₅ - C₆H₄ - C₆H₃ - C₆H₂ - C₆H - C₆

$$C_6H_3 = C_6H_5 - C_6H_4 - C_6H_3 - C_6H_2 - C_6H - C_6$$

3. The general formula for trisubstituted benzene is C₆H₃ = C₆H₅ - C₆H₄ - C₆H₃ - C₆H₂ - C₆H - C₆

4. The general formula for trisubstituted benzene is C₆H₃ = C₆H₅ - C₆H₄ - C₆H₃ - C₆H₂ - C₆H - C₆

5. The general formula for trisubstituted benzene is C₆H₃ = C₆H₅ - C₆H₄ - C₆H₃ - C₆H₂ - C₆H - C₆

6. The general formula for trisubstituted benzene is C₆H₃ = C₆H₅ - C₆H₄ - C₆H₃ - C₆H₂ - C₆H - C₆

7. The general formula for trisubstituted benzene is C₆H₃ = C₆H₅ - C₆H₄ - C₆H₃ - C₆H₂ - C₆H - C₆

$$C_6H_3 = C_6H_5 - C_6H_4 - C_6H_3 - C_6H_2 - C_6H - C_6$$

Section 2: Compounds (Library: Muller (Muller))

Trisubstituted benzene (C₆H₃)

1. The structure of C₆H₃ is a trisubstituted benzene ring.



2. The C₆H₃ bond distances in C₆H₃ indicate delocalized bonding with an average bond length of 1.39 Å, which is shorter than the C-C bond length of 1.54 Å in the case of the C₆H₆ molecule. This is due to the delocalization of the π electrons in the C₆H₃ molecule, which gives it a more stable structure than the C₆H₆ molecule. The C₆H₃ molecule is shown below:



Section 3: Compounds (Library: Muller (Muller))

Structure of Trisubstituted benzene (C₆H₃)



Solution Manual Inorganic Chemistry Huheey

Julie B. Ealy, James Lee Ealy



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