

Solution Manual for General Chemistry 10th Edition by Ebbing

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CHAPTER 2

Atoms, Molecules, and Ions

■ SOLUTIONS TO EXERCISES

Note on significant figures: If the final answer to a solution needs to be rounded off, it is given first with one nonsignificant figure, and the last significant figure is underlined. The final answer is then rounded to the correct number of significant figures. In multistep problems, intermediate answers are given with at least one nonsignificant figure; however, only the final answer has been rounded off.

- 2.1. The element with atomic number 17 (the number of protons in the nucleus) is chlorine, symbol Cl. The mass number is $17 + 18 = 35$. The symbol is $^{35}_{17}\text{Cl}$.
- 2.2. Multiply each isotopic mass by its fractional abundance; then sum:
- $$\begin{array}{rcl} 34.96885 \text{ amu} \times 0.75771 & = & 26.496247 \\ 36.96590 \text{ amu} \times 0.24229 & = & \underline{8.956467} \\ 35.452714 & = & 35.453 \text{ amu} \end{array}$$
- The atomic mass of chlorine is 35.453 amu.
- 2.3.
- Se; Group VIA, Period 4; nonmetal
 - Cs; Group IA, Period 6; metal
 - Fe; Group VIIIB, Period 4; metal
 - Cu; Group IB, Period 4; metal
 - Br; Group VIIA, Period 4; nonmetal
- 2.4. Take as many cations as there are units of charge on the anion and as many anions as there are units of charge on the cation. Two K^+ ions have a total charge of $2+$, and one CrO_4^{2-} ion has a charge of $2-$, giving a net charge of zero. The simplest ratio of K^+ to CrO_4^{2-} is 2:1, and the formula is K_2CrO_4 .

Solution Manual Of General Chemistry Ebbing

**David Bookin, Darrell D. Ebbing, Steven
D. Gammon**



Solution Manual Of General Chemistry Ebbing:

Unveiling the Magic of Words: A Review of "**Solution Manual Of General Chemistry Ebbing**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Solution Manual Of General Chemistry Ebbing**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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