

Student Solution Manual

Student Solution Manual: Skoog et al., *Fundamentals of Analytical Chemistry*, 10e,
© 2022, 978-0-357-45055-0, Chapter 6: Sampling, Standardization, and Calibration

Some of the answers below may differ in format but have the same value as your result. Please check with your instructor if a specific format is desired.

Chapter 6

- 6-1.** A 0.005-g sample of a rock is to be analyzed, and iron is to be determined at the ppm level. Determine the type of analysis and type of constituent.

Answer:

The sample size is in the micro range and the analyte level is in the trace range. Hence, the analysis is a micro analysis of a trace constituent.

- 6-3.** Describe the steps in a sampling operation.

Answer:

Step 1: Identify the population from which the sample is to be drawn.

Step 2: Collect the gross sample.

Step 3: Reduce the gross sample to a laboratory sample, which is a small quantity of homogeneous material.

- 6-5.** The following results were obtained for the determination of calcium in a NIST limestone sample: %CaO = 51.33, 51.22, 51.36, 51.21, and 51.44. Five gross samples were then obtained for a carload of limestone. The average percent CaO values for the gross samples were found to be 48.53, 50.75, 48.60, 48.87, and 50.29. Calculate the relative standard deviation associated with the sampling step.

Solution:

$$s_{\text{tot}}^2 = s_{\text{r}}^2 + s_{\text{m}}^2$$

$$\text{From the NIST sample: } s_{\text{m}}^2 = 0.00947$$

$$\text{From the gross sample: } s_{\text{r}}^2 = 1.073$$

$$s_{\text{r}} = \sqrt{1.07302 - 0.00947} = 1.0313$$

$$\text{The relative standard deviation} = \left(\frac{s_{\text{r}}}{\bar{x}} \right) \times 100\% = \left(\frac{1.0313}{51.312} \right) \times 100\% = 2.01\%$$

- 6-7.** Changes in the method used to coat the tablets in Problem 6-6 lowered the percentage of rejects from 8.0% to 3.0%. How many tablets should be taken for inspection if the permissible relative standard deviation in the measurement is to be

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Rose Army



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