

Solutions, acids, and bases

1. VOCABULARY REVIEW OF SOLUTIONS

Solute - what is **dissolved** in the solvent to form a solution.

Solvent - the substance that **dissolves** the solute, ordinarily a liquid when the solute and the solvent come together to form a **homogeneous mixture**.

Miscible - two mix in any proportions (have to be mixed) - liquids will pour each over

Immiscible - don't mix in any proportions (have to be mixed)

Examples of miscible-immiscible - ethanol-water, oil-water, oil-water, oil-water

Solute - **amount** of **being dissolved** in liquid

Insoluble - a substance (tends to be a solid) that **will not dissolve** in a solvent - an example is sand and water

Suspension - looks like a **solution** when **stirred** but **settles** a bit of water (suspensions) also it is a mixture - gaseous mixture where the small solid particles of a substance are spread throughout a liquid without settling.

Unsaturated - **solids** with **solubility chart** when the temperature would be enough to make a substance dissolve but at a greater amount of **solute** - **solubility less** - **solute** **exceeds** what can be dissolved (at that temp)

Saturated - **solids** with **solubility chart** when the temperature - **solubility** relationship is at the line in other words a **solution** with **solute** has **reached** at the **maximum possible amount**.

Supersaturated - contains **more** than the **max. amount** of **solute** that can be dissolved at the temperature - on the solubility chart this would mean the grams of solute would go above the line.

2. SOLUBILITY VS. TEMPERATURE CHART



Above the line - **supersaturated**

At the line - **saturated**

Under the line - **unsaturated**

3. METHODS OF SEPARATION

Filtration - separate particles by **size**

Distillation - separate by **boiling point** - **more** and **liquids** to be separated.

Decantation - allows one to remove a **liquid** from a **solid** by **transferring** the **liquid** to a **different** **beaker**.

Centrifugation - separating molecules by their **densities** by **spinning** them **quickly**.

Decanting - process of separating **insoluble** or **suspension** solutions by **pouring** away the **liquid** from the **solid**.

Chromatography - separation by the **rates** at which components of different solutions **move** through a **medium** (e.g. gel electrophoresis)

Solutions Acids And Bases Study Guide

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Starkey, 2025-03-18 Success in organic chemistry requires mastery in two core aspects fundamental concepts and the skills needed to apply those concepts and solve problems With Organic Chemistry Student Study Guide and Solutions Manual 5th Edition students can learn to become proficient at approaching new situations methodically based on a repertoire of skills These skills are vital for successful problem solving in organic chemistry *The Complete Chemistry Study Guide and Note Cards and MCAT* Konstantinos Papadopoulos, 2012-07-06 *Student Study Guide/Solutions Manual for Essentials of General, Organic, and Biochemistry* Denise Guinn, Rebecca Brewer, Rachel C. Lum, 2009-09-15 The Student Study Guide and

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