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L. C. Forier (ed)



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$$\text{molality} = \frac{\text{moles solute}}{\text{kilograms solvent}}$$
 mathematical manipulation of molality is the

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$$\text{begin array l} \frac{\text{mass of a component}}{\text{mass of component in solution total mass of solution}} \times 100 \text{ end array}$$

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