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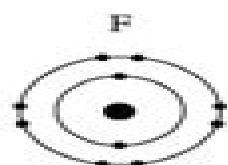
1 How many electrons are present in the following elements?

- a** Carbon ____ **b** Chlorine ____ **c** Indium ____ **d** Calcium ____
e Tungsten ____ **f** Antimony ____ **g** Titanium ____ **h** Tin ____

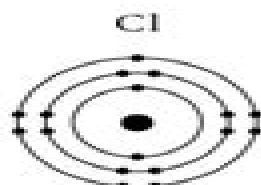
2 Using the periodic table, determine the number of valance electrons for the elements listed below.

- a** Copper ____ **b** Xenon ____ **c** Hydrogen ____ **d** Aluminum ____
e Barium ____ **f** Iodine ____ **g** Carbon ____ **h** Phosphorous ____

3 Circle the valence electrons only. What is the atomic number?



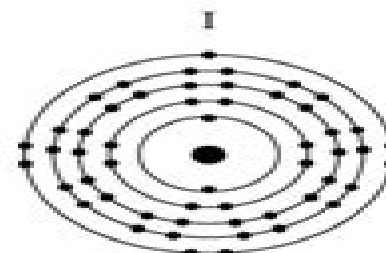
Atomic number(Z) =



Atomic number(Z) =



Atomic number(Z) =



Atomic number(Z) =

4 Fill in the table with correct values.

Element	Ion symbol	Number of Protons	Number of electrons	Charge	Ion type
Fluorine	F ⁻			-1	Anion
		53	54		
		16		-2	
Potassium				+1	
Calcium					

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