

1/5/16

Mole Ratio Worksheet

Key

① Given this equation: $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$, write the following molar ratios:

a) N_2 / H_2 1:3

b) N_2 / NH_3 1:2

c) H_2 / NH_3 3:2

② Given the following equation: $8\text{H}_2 + \text{S}_8 \rightarrow 8\text{H}_2\text{S}$, write the following molar ratios:

a) $\text{H}_2 / \text{H}_2\text{S}$ 8:8 \rightarrow 1:1 (reduced)

b) H_2 / S_8 8:1

c) $\text{H}_2\text{S} / \text{S}_8$ 8:1

③ Answer the following questions for this equation: $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$

a) What is the $\text{H}_2 / \text{H}_2\text{O}$ molar ratio? 2:2 \rightarrow 1:1 (reduced)

b) Suppose you had 20 moles of H_2 on hand and plenty of O_2 , how many moles of H_2O could you make? $\frac{20 \text{ mole H}_2}{2 \text{ H}_2} \times \frac{2 \text{ H}_2\text{O}}{1} = \boxed{20 \text{ mole H}_2\text{O}}$

c) What is the $\text{O}_2 / \text{H}_2\text{O}$ molar ratio?

1:2

d) Suppose you had 20 moles of O_2 and enough H_2 , how many moles of H_2O could you make? $\frac{20 \text{ mol O}_2}{1 \text{ O}_2} \times \frac{2 \text{ H}_2\text{O}}{1} = \boxed{40 \text{ mol H}_2\text{O}}$

4) Use this equation: $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$, for the following problems

a) If you used 1 mole of N_2 , how many moles of NH_3 could be produced? $\boxed{2}$

b) If 10 moles of NH_3 were produced, how many moles of N_2 would be required? $\frac{10 \text{ mol NH}_3}{2 \text{ NH}_3} \times \frac{1 \text{ N}_2}{1} = \boxed{5 \text{ mol N}_2}$

c) If 3.00 moles of H_2 were used, how many moles of NH_3 would be made? $\frac{3 \text{ mol H}_2}{3 \text{ H}_2} \times \frac{2 \text{ NH}_3}{1} = \boxed{2 \text{ mol NH}_3}$

d) If 0.600 moles of NH_3 were produced, how many moles of H_2 are required? $\frac{0.6 \text{ mol NH}_3}{2 \text{ NH}_3} \times \frac{3 \text{ H}_2}{1} = \boxed{0.9 \text{ mol H}_2}$

What Is The Mole Ratio Worksheet Answers

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Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998

Chemical Process Engineering Volume 1 Rahmat Sotudeh-Gharebagh, A. Kayode Coker, 2022-05-03 Written by two of the most prolific and respected chemical engineers in the world this groundbreaking two volume set is the new standard in the industry offering engineers and students alike the most up to date comprehensive and state of the art coverage of processes and best practices in the field today This first new volume in a two volume set explores and describes integrating new tools for engineering education and practice for better utilization of the existing knowledge on process design Useful not only for students professors scientists and practitioners especially process chemical mechanical and metallurgical engineers it is also a valuable reference for other engineers consultants technicians and scientists concerned about various aspects of industrial design The text can be considered as a complementary text to process design for senior and graduate students as well as a hands on reference work or refresher for engineers at entry level The contents of the book can also be taught in intensive workshops in the oil gas petrochemical biochemical and process industries The book provides a detailed description and hands on experience on process design in chemical engineering and it is an integrated text that focuses on practical design with new tools such as Excel spreadsheets and UniSim simulation software Written by two industry and university s most trustworthy and well known authors this book is the new standard in chemical biochemical pharmaceutical petrochemical and petroleum refining Covering design analysis simulation integration and perhaps most importantly the practical application of Microsoft Excel UniSim software this is the most comprehensive and up to date coverage of all of the latest developments in the industry It is a must have for any engineer or student s library

Improving Student Comprehension of Stoichiometric Concepts Connie Lynn Bannick Kemner, 2007

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Table of Contents What Is The Mole Ratio Worksheet Answers

1. Understanding the eBook What Is The Mole Ratio Worksheet Answers
 - The Rise of Digital Reading What Is The Mole Ratio Worksheet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying What Is The Mole Ratio Worksheet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an What Is The Mole Ratio Worksheet Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from What Is The Mole Ratio Worksheet Answers
 - Personalized Recommendations
 - What Is The Mole Ratio Worksheet Answers User Reviews and Ratings
 - What Is The Mole Ratio Worksheet Answers and Bestseller Lists
5. Accessing What Is The Mole Ratio Worksheet Answers Free and Paid eBooks
 - What Is The Mole Ratio Worksheet Answers Public Domain eBooks
 - What Is The Mole Ratio Worksheet Answers eBook Subscription Services
 - What Is The Mole Ratio Worksheet Answers Budget-Friendly Options
6. Navigating What Is The Mole Ratio Worksheet Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - What Is The Mole Ratio Worksheet Answers Compatibility with Devices
 - What Is The Mole Ratio Worksheet Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of What Is The Mole Ratio Worksheet Answers
 - Highlighting and Note-Taking What Is The Mole Ratio Worksheet Answers
 - Interactive Elements What Is The Mole Ratio Worksheet Answers
8. Staying Engaged with What Is The Mole Ratio Worksheet Answers

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers What Is The Mole Ratio Worksheet Answers
- 9. Balancing eBooks and Physical Books What Is The Mole Ratio Worksheet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection What Is The Mole Ratio Worksheet Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine What Is The Mole Ratio Worksheet Answers
 - Setting Reading Goals What Is The Mole Ratio Worksheet Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of What Is The Mole Ratio Worksheet Answers
 - Fact-Checking eBook Content of What Is The Mole Ratio Worksheet Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
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 - Interactive and Gamified eBooks

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