

Name: _____

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Titration Practice Worksheet

Find the requested quantities in the following problems:

- 1) Write a balanced equation for the reaction between NaOH and HCl. Use this equation to answer problems 2-4

- 2) If it takes 54 mL of 0.1 M NaOH to neutralize 125 mL of an HCl solution, what is the concentration of the HCl?

- 3) If it takes 25 mL of 0.05 M HCl to neutralize 345 mL of NaOH solution, what is the concentration of the NaOH solution?

- 4) A 25.0 mL sample of HCl was titrated to the endpoint with 15.0 mL of 2.0 M NaOH. What is the molarity of HCl?

- 5) Write a balanced equation for the reaction between Ca(OH)_2 and HNO_3 . Use this equation to answer problems 6 -8.

- 6) If it takes 50.0 mL of 0.5 M Ca(OH)_2 solution to completely neutralize 125 mL of HNO_3 solution, what is the concentration of the HNO_3 solution?

- 7) How many mL of 0.50 M HNO_3 is necessary to titrate 25.0 mL of 0.05 M Ca(OH)_2 solution to the endpoint?

- 8) If it takes 75.0 mL of 1.5 M HNO_3 solution to completely neutralize 125 mL of Ca(OH)_2 , what is the concentration of the Ca(OH)_2 solution?

Titration Worksheet Chemfiesta

Scott C. Dulebohn

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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