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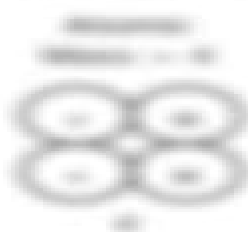
## ATOMIC NUMBERS, PROTONS, NEUTRONS AND ELECTRONS

The number of protons in each atom of an element is its atomic number. The atomic number is also the number of electrons since atoms themselves are electronically neutral. Because neutrons and protons are almost the same mass, the total number of protons and neutrons in an atom is the atomic mass. Therefore, by subtracting the atomic number from the atomic mass you can calculate the number of neutrons.

This is why the mass number and atomic number, but not the number of neutrons, is shown in the periodic table. You can use the information provided below to determine the missing values in the chart below.



Example: Helium atom



Use this knowledge and the information provided below to determine the missing values in the chart below.

Element	Symbol	Atomic Number	Atomic Mass	Protons	Neutrons	Electrons
Hydrogen	H					
Helium	He					
Lithium	Li					
Boron	B					
Carbon	C					
Nitrogen	N					
Oxygen	O					
Fluorine	F					
Neon	Ne					
Sodium	Na					
Magnesium	Mg					
Aluminum	Al					
Silicon	Si					
Phosphorus	P					
Sulfur	S					
Chlorine	Cl					
Argon	Ar					
Potassium	K					
Calcium	Ca					

# Where Are The Electrons Answer Key

**Shyam Wuppuluri, Ian Stewart**



## **Where Are The Electrons Answer Key:**

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