

## Chapter Test A

### A. Completion

Directions: Complete each blank with the appropriate word or phrase.

1. It is composed of two hydrogen atoms and one oxygen atom, making water a covalent compound.
2. It is necessary to add energy to boil water, to convert it from a liquid to a gas.
3. It is known that water is a universal solvent because of its polarity.
4. It is necessary to add energy to boil water, to convert it from a liquid to a gas.
5. It is known that water is a universal solvent because of its polarity.
6. It is known that water is a universal solvent because of its polarity.
7. It is known that water is a universal solvent because of its polarity.
8. It is known that water is a universal solvent because of its polarity.
9. It is known that water is a universal solvent because of its polarity.
10. It is known that water is a universal solvent because of its polarity.
11. It is known that water is a universal solvent because of its polarity.
12. It is known that water is a universal solvent because of its polarity.
13. It is known that water is a universal solvent because of its polarity.
14. It is known that water is a universal solvent because of its polarity.
15. It is known that water is a universal solvent because of its polarity.

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|-----|-------|
| 1.  | _____ |
| 2.  | _____ |
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| 4.  | _____ |
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| 9.  | _____ |
| 10. | _____ |
| 11. | _____ |
| 12. | _____ |
| 13. | _____ |
| 14. | _____ |
| 15. | _____ |

### B. Multiple Choice

Directions: Choose the best answer for each question.

16. Which of the following is not a property of water?
  - a. high heat capacity
  - b. low surface tension
  - c. high boiling point
  - d. high density of solid form

# Water And Aqueous Systems Chapter Test A Answers

**Lingsheng Yao**



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