

Name: _____ Period: _____

Properties of Water Worksheet

Word Bank:

covalent
tension
solvent
polar

dissolve
cohesion
negatively
surface

adhesion
positively
universal

1. The hydrogen and oxygen atoms are held together by _____ bonds.
2. The electrons are not shared equally creating a _____ molecule.
3. The polarity of water allows it to _____ most substances. Because of this, it is referred to as the _____.
4. Water molecules stick to other water molecules. This property is called _____.
5. Hydrogen bonds form between adjacent water molecules because the _____ charged hydrogen end of one water molecule attracts the _____ charged oxygen end of another water molecule.
6. Water molecules stick to other materials due to its polar nature. This property is called _____.
7. _____ creates the skin-like surface formed due to the polar nature of water.
8. Circle the letter of each sentence that is true about water's structure.
 - a. Water is made up of atoms bonded to form molecules.
 - b. Water contains half as many hydrogen atoms as oxygen atoms.
 - c. Water molecules tend to push away from each other.
 - d. The chemical formula for water is H_2O .
9. One side of the water molecule has a positive charge, while the other side has a negative charge. What do the charges indicate about the molecule?
 - a. Water is a polar molecule.
 - b. Water is a nonpolar molecule
 - c. Water is an ionic compound
 - d. Water is an ion
10. True or false: Hydrogen bonds are strong and require a lot of energy to break.
11. True or false: A substance that hates or fears water is hydrophobic.

Water And Aqueous Systems Worksheets

**Henry Lutz Ehrlich,Dianne K.
Newman,Andreas Kappler**



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Chemical Resistance Data Sheets Rapra Technology Limited, 1993 Over 190 chemicals are assessed for 44 different plastics

Handbook on Material and Energy Balance Calculations in Material Processing Arthur E. Morris, Gordon Geiger, H. Alan Fine, 2012-01-03 Lately there has been a renewed push to minimize the waste of materials and energy that accompany the production and processing of various materials This third edition of this reference emphasizes the fundamental principles of the conservation of mass and energy and their consequences as they relate to materials and energy New to this edition are numerous worked examples illustrating conventional and novel problem solving techniques in applications such as semiconductor processing environmental engineering the production and processing of advanced and exotic materials for aerospace electronic and structural applications

Environmental and Low-Temperature Geochemistry Peter Ryan, 2019-12-16 Environmental and Low Temperature Geochemistry presents conceptual and quantitative principles of geochemistry in order to foster understanding of natural processes at and near the earth's surface as well as anthropogenic impacts and remediation strategies It provides the reader with principles that allow prediction of concentration speciation mobility and reactivity of elements and compounds in soils waters sediments and air drawing attention to both thermodynamic and kinetic controls The scope includes atmosphere terrestrial waters marine waters soils sediments and rocks in the shallow crust the temporal scale is present to Precambrian and the spatial scale is nanometers to local regional and global This second edition of Environmental and Low Temperature Geochemistry provides the most up to date status of the carbon cycle and global warming including carbon sources sinks fluxes and consequences as well as emerging evidence for and effects of ocean acidification Understanding environmental problems like this requires knowledge based in fundamental principles of equilibrium kinetics basic laws of chemistry and physics empirical evidence examples from the geological record and identification of system fluxes and reservoirs that allow us to conceptualize and understand This edition aims to do that with clear explanations of fundamental principles of geochemistry as well as information and approaches that provide the student or researcher with knowledge to address pressing questions in environmental and geological sciences New content in this edition includes Focus Boxes one every two or three pages providing case study examples e g methyl isocyanate in Bhopal origins and health effects of asbestiform minerals concise explanations of fundamental concepts e g balancing chemical equations isotopic fractionation using the K_{eq} to predict reactivity and useful information e g units of concentration titrating to determine alkalinity measuring redox potential of natural waters Sections on emerging contaminants for which knowledge is rapidly increasing e g perfluorinated compounds pharmaceuticals and other domestic and industrial chemicals Greater attention to interrelationships of inorganic organic and biotic phases and processes Descriptions theoretical frameworks and examples of emerging methodologies in geochemistry research e g clumped C O isotopes to assess seawater temperature over geological time metal stable isotopes to assess source and

transport processes X ray absorption spectroscopy to study oxidation state and valence configuration of atoms and molecules Additional end of chapter problems including more quantitatively based questions Two detailed case studies that examine fate and transport of organic contaminants VOCs PFCs with data and interpretations presented separately These examples consider the chemical and mineralogical composition of rocks soils and waters in the affected system microbial influence on the decomposition of organic compounds the effect of reduction oxidation on transport of Fe As and Mn stable isotopes and synthetic compounds as tracers of flow geological factors that influence flow and implications for remediation The interdisciplinary approach and range of topics including environmental contamination of air water and soil as well as the processes that affect both natural and anthropogenic systems make it well suited for environmental geochemistry courses at universities as well as liberal arts colleges

Water and Ions in Biomolecular Systems D. Vasilescu, 2013-03-08 *Journal of Research of the National Bureau of Standards* United States. National Bureau of Standards, 1960

Advances in Liquid Crystals Glenn H. Brown, 2013-10-22 *Advances in Liquid Crystals Volume I* is a collection of papers dealing with lyotropic and thermotropic liquid crystals One paper discusses lyotropic liquid crystalline phases in systems of amphiphilic compounds including the liquid crystalline phases lyotropic liquid crystalline phase the gel phases and the transition between normal and reversed mesophases Another paper describes the ordering and structure of liquid crystals by reviewing research work on X ray analysis of thermotropic liquid crystals One paper explains the studies made on the mesomorphic behavior of block copolymers including the incompatibility segregation and organization of polymers The paper notes that plastic crystals and liquid crystals are different manifestations of the duality of the melting process in molecular crystals resulting in the loss of positional and orientation order One paper describes the essential linear defects found in solid crystals which have been researched in terms of the major role they play in macroscopic properties particularly plasticity as the defects or dislocations of translation G Friedel's work classifies liquid crystals into an intermediary symmetry composed of the nematic the cholesteric and the smectic phases This intermediary symmetry occurs between the liquid type symmetry and the solid type symmetry This collection can prove useful for researchers in liquid crystals and industrial scientists in the field of chemistry and polymer composites

Chemical Safety Sheets Chemical Industry Association, VNCI, the Netherlands, 2012-12-06
Water-resources Investigations Report, 1996

Hydrology, Water Quality, and Effects of Drought in Monroe County, Michigan J. R. Nicholas, 1996

Water Pollution Research Great Britain. Water Pollution Research Board, 1928

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