#### Study Guide for Chemistry EOC Assessment

Your test will consist of multiple-choice questions over the following topics.

- 1. Metric units for mass, length, volume
- Metric to metric conversions (K H Da m/l/g D C M)
- Know how to put numbers in scientific notation and take them out.
- 4. Significant Digit rules
- Difference between chemical and physical properties and changes. There will be multiple questions on this.
- 6. Ways to tell if a chemical change (chemical reaction) is taking place.
- 7. Pure substances and mixtures
  - a. heterogeneous mixture
  - b. homogeneous mixture
- 8. Density & how it relates to floating and sinking
- Use the equation for density
- Density = Mass/volume (g/ml) (g) (ml)
- Know the phases of matter and the characteristics of each phase.
- 11. Law of Conservation of Matter
- 12. What is a mole in chemistry?
- 13. Convert grams to moles, and moles to grams
- Define an atom
- 15. Structure of an atom and where each subatomic particle is located.
  - a. Nucleus (contains protons and neutrons)
  - b. Protons (positive)
  - c. Neutrons (neutral)
  - d. Electrons (negative)
- 16. Isotopes
- 17. Elements
- 18. If given atomic # and mass, find the number of protons and neutrons in an atom.
- 19. Find the number of valence electrons, and tell what it says about reactivity.
- 20. Periodic Table
  - a. Know about the families
  - b. Group vs. period
- 21. Know about the properties of metals and nonmetals.
  - a. Ductile, malleable, conductors
- 22. Ionic & Covalent Compounds

How do they form? (Transfer or gain electrons) Characteristics of Each

- Be able to name ionic compounds
- 24. Predict oxidation numbers
- 25. Chemical Equations:

Know the parts of an equation

5 main types of reactions-know about them and be able to identify them (Synthesis, Decomposition, Single Displacement, Double Displacement, Combustion)

26. Acids & Bases-

Macro and Microscopic properties of each

What ions do they form, how do they affect litmus, what is their pH range?

Identify examples of each- Mono, di, and polyprotic acids

Tell if they are strong or weak

Acid-base indicator

# **Study Guide For Chemistry Eoc**

Seager/Slabaugh

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Cracking the Virginia SOL Tina Walton, Princeton Review (Firm), 2000 The Princeton Review realizes that acing the English Reading Literature and Research exam is very different from getting straight As in school The Princeton Review doesn't try to teach students everything there is to know about English only the techniques they ll need to score higher on the exam There's a big difference In Cracking the Virginia SOL EOC English Reading Literature and Research The Princeton Review will teach test takers how to think like the test makers and Eliminate incorrect answer choices by using Process of Elimination and other techniques Familiarize students with every type of reading material that might be on the test Use the two pass system to get a grip on the test structure and raise scores Hone individual skills with practice questions for each type of test question This book includes 2 full length simulated exams All of TPR s sample test questions are just like the ones test takers will see on the actual End of Course English Reading Literature and Research exam and TPR fully explains every solution Contents Include The Mystery Test Structure and Strategies II Terms Literary Movements and Writers Literature Review III Literary Elements and Printed and Resource Materials The Elements of Literature A Variety of Printed Material A South Dakota Chemistry Eoc Success Strategies Variety of Resource Material IV The Princeton Review Practice Tests Study Guide South Dakota Eoc Exam Secrets Test Prep, 2014-08-22 South Dakota Chemistry EOC Success Strategies helps you ace the South Dakota End of Course Exams without weeks and months of endless studying Our comprehensive South Dakota Chemistry EOC Success Strategies study guide is written by our exam experts who painstakingly researched every topic and concept that you need to know to ace your test Our original research reveals specific weaknesses that you can exploit to increase your exam score more than you ve ever imagined South Dakota Chemistry EOC Success Strategies includes The 5 Secret Keys to South Dakota EOC Success Time is Your Greatest Enemy Guessing is Not Guesswork Practice Smarter Not Harder Prepare Don t Procrastinate Test Yourself A comprehensive General Strategy review including Make Predictions Answer the Question Benchmark Valid Information Avoid Fact Traps Milk the Question The Trap of Familiarity Eliminate Answers Tough Questions Brainstorm Read Carefully Face Value Prefixes Hedge Phrases Switchback Words New Information Time Management Contextual Clues Don t Panic Pace Yourself Answer Selection Check Your Work Beware of Directly Quoted Answers Slang Extreme Statements Answer Choice Families Along with a complete in depth study guide for your specific South Dakota EOC exam and much more Study Guide and Solutions Manual for Seager/Slabaugh's Chemistry for Today Seager/Slabaugh, 2004 The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even numbered end of chapter exercises New additional activities have been added to include a review of each section of the chapter and a section entitled Tying It All Together with a Laboratory Application **STAAR EOC Chemistry Assessment Secrets Study Guide** Mometrix Media LLC,2016-10-24 Includes Practice Test Questions

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Introduction to Organic Chemistry William H. Brown, Thomas Poon, 2016-01-13 Introduction to Organic Chemistry 6th Edition provides an introduction to organic chemistry for students who require the fundamentals of organic chemistry as a requirement for their major It is most suited for a one semester organic chemistry course In an attempt to highlight the

relevance of the material to students the authors place a strong emphasis on showing the interrelationship between organic chemistry and other areas of science particularly the biological and health sciences. The text illustrates the use of organic chemistry as a tool in these sciences it also stresses the organic compounds both natural and synthetic that surround us in everyday life in pharmaceuticals plastics fibers agrochemicals surface coatings toiletry preparations and cosmetics food additives adhesives and elastomers. This text is an unbound three hole punched version Access to WileyPLUS sold separately

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been revised with an even sharper focus on practical skills and procedures The first half of the book is devoted to safe laboratory practice as well as purification and analytical techniques particularly spectroscopic analysis The second half contains step by step experimental procedures each one illustrating a basic principle or important reaction type Tried and tested over almost three decades over 100 validated experiments are graded according to their complexity and all are chosen to highlight important chemical transformations and to teach key experimental skills New sections cover updated health and safety guidelines additional spectroscopic techniques electronic notebooks and record keeping and techniques such as semi automated chromatography and enabling technologies such as the use of microwave and flow chemistry New experiments include transition metal catalysed cross coupling organocatalysis asymmetric synthesis flow chemistry and microwave assisted synthesis Key aspects of this third edition include Detailed descriptions of the correct use of common apparatus used in the organic laboratory Outlines of practical skills that all chemistry students must learn Highlights of aspects of health and safety in the laboratory both in the first section and throughout the experimental procedures Four new sections reflecting advances in techniques and technologies from electronic databases and information retrieval to semi automated chromatography More than 100 validated experiments of graded complexity from introductory to research level A user friendly experiment directory An instructor manual and PowerPoint slides of the figures in the book available on a companion website A comprehensive guide to contemporary organic chemistry laboratory principles procedures protocols tools and techniques Experimental Organic Chemistry Third Edition is both an essential laboratory textbook for students of chemistry at all levels and a handy bench reference for experienced chemists Chemistry Exam Study Guide Houk, 2018-12-03 EVERYTHING YOU NEED TO HELP SCORE A PERFECT SCORE Equip yourself to ace your Chemistry Exam with this comprehensive study guide including full length practice tests thorough content reviews targeted strategies for every section Books In Print 2004-2005 Ed Bowker Staff, Staff Bowker, Ed, 2004

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