

Name _____ Date _____ Class _____

CHAPTER 15 STUDY GUIDE FOR CONTENT MASTERY

Solutions

Section 15.1 What are solutions?

In your textbook, read about the characteristics of solutions.

Use each of the terms below just once to complete the passage.

immiscible	liquid	solution	solute	solution
insoluble	miscible	solute	solute	solvent

Air is air! (1) solution of oxygen gas dissolved in nitrogen gas. The oxygen in air is the (2) solute and nitrogen is the (3) solvent. Because oxygen gas dissolves in a solvent, oxygen gas is air! (4) solute substances. A substance that does not dissolve is (5) insoluble. (6) liquid solutions are the most common type of solutions. If one liquid is soluble in another liquid, such as acetic acid in water, the two liquids are (7) miscible. However, if one liquid is insoluble in another, the liquids are (8) immiscible.

Read about solvation in aqueous solutions in your textbook.

The diagram shows the hydration of solid sodium chloride to form an aqueous solution. Use the diagram to answer the following questions.



8. Hydration is solvation in which the solvent is water. What is solvation?

Solution is the process of surrounding solute particles with solvent particles to form a solution.

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CHAPTER 15 STUDY GUIDE FOR CONTENT MASTERY

Section 15.1 continued

10. As sodium chloride dissolves in water, what happens to the sodium and chloride ions?

The sodium and chloride ions are separated and surrounded by the water molecules.

11. Explain the attraction of the water molecules around the sodium ions and chloride ions.

Because the sodium ion is positively charged, it attracts the negatively charged portion of the water molecule (the oxygen atom) and repels the positively charged portion of the water molecule (the hydrogen atoms). Because the chloride ion is negatively charged, it attracts the positively charged portion of the water molecule and repels the negatively charged portion.

12. How does the strength of the attraction between water molecules and sodium and chloride ions compare with the strength of the attraction between the sodium ions and chloride ions? How do you know?

The attraction between the water molecules and the sodium and chloride ions is greater than the attraction between the sodium and chloride ions. The greater strength of attraction between the water molecules and the ions is what causes the solvation process to occur.

13. List three ways that the rate of solvation may be increased.

stirring or shaking the solution, breaking the solute into smaller pieces, and heating the solvent

In your textbook, read about heat of solution, solubility, and factors that affect solubility.

For each statement below, write true or false.

True _____ 14. The overall energy change that occurs when a solution forms is called the heat of solution.

False _____ 15. Solubility is a measure of the maximum amount of solute that dissolves in a given amount of solvent at a specified temperature and pressure.

False _____ 16. Solvation continues as long as the solvation rate is less than the crystallization rate.

True _____ 17. In a saturated solution, solvation and crystallization are in equilibrium.

True _____ 18. Additional solute can be dissolved in an unsaturated solution.

True _____ 19. The solubility of a gas dissolved in a liquid decreases as the temperature of the solution increases.

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Robert R. Redfield



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Handbook of Research on E-Learning Applications for Career and Technical Education: Technologies for Vocational Training Wang, Victor X.,2009-05-31 Provides an authoritative reference collection on leading international insights into the integration of technology tools and applications with adult and vocational instruction Deep Learning in Adaptive Learning: Educational Behavior and Strategy Chia-Chen Chen,Patrick Hung,Erol Eğrioğlu,Kuo-Lun Hsiao,2022-09-14 *The Education Index* ,1984 School Choice Or Best Systems Margaret C Wang,Herbert J. Walberg,2001-03-01 This book addresses one of the most urgent questions in American society today one that is currently in the spotlight and hotly debated on all sides Who shall rule the schools parents or educators School Choice or Best Systems What Improves Education presents an overview of research and practical applications of innovative even radical school reforms being implemented across the United States These fall along a continuum ranging from parental choice to best systems At the one extreme are schools of choice which allow parents to choose and even govern schools for their children These include charter schools traditional private and parochial schools schools that are privately governed but publicly funded through vouchers and those that are funded by private scholarships provided by both corporations and wealthy individuals At the other extreme are centralized state or district systems based on reform initiatives and new systems of education that have been developed in response to views of citizens and legislators that schools can do much better These schools which specify uniform goals policies and programs for each school are highly innovative systems based on research or representing advanced thinking about what works and have attracted wide interest Important questions related to schools of choice and best systems are addressed How can we choose among schools of choice and best systems Among the various approaches within each of these alternatives How can we understand their guiding principles and operational practices What results do they produce How can we evaluate their claims In choosing among the alternatives how should issues of student achievement accountability costs feasibility and equity be factored in This volume brings together leading researchers and education leaders who have carried out the latest studies and advances in the field providing a forum for them to set forth the arguments and evidence that will be most helpful in making choices for tomorrow s schools It does not provide a single right answer values and preferences differ across parents schools districts and states However there are benefits for all from seeing the rigorous research challenging thinking and alternate points of view this volume presents *Executive Skills and Reading Comprehension* Kelly B. Cartwright,2023-02-02 How do K 12 students become self regulated learners who actively deploy comprehension strategies to make meaning from texts This cutting edge guide is the first book to highlight the importance of executive skills for improving reading comprehension Chapters review the research base for particular executive functions m such as planning organization cognitive flexibility and impulse control m and present practical skills building strategies for the classroom Detailed examples show what each skill looks like in real readers and sidebars draw

explicit connections to the Common Core State Standards CCSS **Technological Transformation of the Indian Education Pedagogy, Policy , And Progress** Dr. Abdul Salam Khan , **ICSSIET CONGRESS 3st International Congress on Social Sciences, Innovation and Educational Technologies PROCEEDINGS BOOK** Aynura Valiyeva & Thomas Basil John,Jahirul Islam & Md. Harun Rashid,Dr. Gökhan Sümer,Ergün Yurtbakan & Durmuş Ekiz,Hilal Dilan Bayram & Assoc. Prof. Derya Girgin,Dr.Öğr.Üyesi Ümit Şevik,İpeksu Sözüpek & Assoc. Prof. Derya Girgin,Assist. Prof. Hakan Ulum,Emil Raul oğlu Ağayev,Kadir Gökoğlan & Nazan Güngör Karyağdı,Lecturer, Dr. Asiye Yüksel & Lecturer, Dr. Barış Demir,Prof. Dr.D.Rajasekar, Dr.S.R.Deepa,Divyaranjani, Madan Mohan.S,Assist. Prof. Divyaranjani, B.Krishnaveni,Assist. Prof. Mrs. R.Divyaranjani & Mr.Sreerag R.G, Student,Mutia Ismail, Ibnu Austrindanney Sina Azhar , Iskandar Muda , Luigi Pio Leonardo Cavaliere , Andre Giovanni Febu Tamba, Verchiana Isnata,Azhar Maksum, Ibnu Austrindanney Sina Azhar, Iskandar Muda, Luigi Pio Leonardo Cavaliere , Nifta Salsabilla, Tiara Surahva,Iskandar Muda , Erlina, Luigi Pio Leonardo Cavaliere , Siti Arla Anita Suheri Hasibuan, Karina Maharani-,Hasan Basri Tarmizi, Azhar Maksum, Iskandar Muda , Luigi Pio Leonardo Cavaliere, Gustiandi, Muhammad Hirzi,Novriest Umbu Walangara Nau, Luigi Pio Leonardo Cavaliere, Kristena Esther Angkow, Vicky Joshua,Chikezie Kalu, Olani Bekele, Simeon Ebhota,Assist. Prof. Saumyabrata Nath Dr. Manisha Choudhary Dr. Sonali N. Tholia Miss Gayatri R. Pillai,Mellouli Amal & Tijani Omar,Dr. Shams Ur Rahman, Mr. Jawad Kabir, Mr. Luigi Pio Leonardo Cavaliere, Dr. Afef Khalil Afraseyab Khattak,Dr. Ijaz Yusuf and Manahil Fatima,Quyen Le Thi To, Duyen Nguyen Thi My, Dung Phuong Le,Ajeet Jaiswal,Dr. Velankanni Alex, Dr. Ferdin Joe John Joseph, Mr. Luigi Pio Leonardo Cavaliere,Mina Ghiasabadi Farahani,Prince Verma,Eze Benneth, PhD,H.G. Hasanov, I.M. Zeynalov,Mr. S. Arunkumar & Jipson Joseph,Dr. A. Shameem & Mohammed Ismail P.,Dr. A. Shameem,Jahirul Islam,Putri Hergianasari, Rizki Amalia Yanuarta, Mr. Luigi Pio Leonardo Cavaliere,Dr. Velankanni Alex- Luigi Pio Leonardo Cavaliere,Assist. Prof. Mr. Sagar Bhatt,Gulshan Sadaf, Maimona Saleem, Fakhira Zaman,Md. Harun Rashid, Wang Hui,Luigi Pio Leonardo Cavaliere, Dr. Valbona Çinaj Ribaj, Melania Riefolo, Luigi Blescia,Medynska Nataliia, Hunko Liudmyla,Javiera Rosell, Feridun Kaya & Alvaro Vergés,Marco I. Bonelli,Lian Junxiang,Duong Anh Son, Tran Vang-Phu & Luu Hoang Dzung,Monica Mastrantonio, Raveenthiran Vivekanantharasa Raveenthiran Vivekanantharasa & Lidia Martinez,Shilpi Sharma & Rana Majumdar,Esra S. Döngül,Pham Yen Nhi,2022-09-10 ICSSIET CONGRESS 3st International Congress on Social Sciences Innovation and Educational Technologies PROCEEDINGS BOOK **The Oxford Handbook of Language and Social Psychology** Thomas Holtgraves,2014 The Oxford Handbook of Language and Social Psychology is a unique and innovative compilation of research that lies at the intersection of language and social psychology Contributors address the role of social processes in language the linguistic underpinnings of social psychological processes the creation of meaning and the important role played by language and social psychology in applied topics **The China Weekly Review** ,1933-06 **The Technology Teacher** ,1975 Millard's Review of the Far East ,1933 Vol 34 includes Special tariff conference issue Nov 6 1925

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