

Name: _____

Atomic Structure & the Periodic Table Study Guide

Atomic Structure

Particle	Charge	Where are they found?
Proton	Positive	Nucleus
Neutron	Neutral/ no charge	Nucleus
Electron	negative	Electron cloud/energy levels outside nucleus

1. What is the rule for writing chemical symbols like Na or Ca? **Capitalize the first letter; lower case second letter**
2. What are the vertical columns on the periodic table called? Groups or families
 - a. How many of these are on the periodic table? **18**
 - b. Why are the elements in these columns? They have the same chemical properties (& same # of valence electrons)
 - c. Elements in the same group have the same number of valence electrons.
3. What are the horizontal rows on the periodic table called? periods
 - a. How many of these are on the periodic table? **7**
 - b. Elements in these rows have the same number of energy (*electron shells*) levels.
4. What are valence electrons? **Electrons in the outer energy level**
5. How many electrons are in each energy level?
1st energy level **2** 2nd energy level **8** 3rd energy level **8**
6. The atomic number is equal to the # of protons and also # of electrons
7. The atomic mass is equal to the # protons + # of neutrons
8. How do you find the number of neutrons? Atomic mass – atomic number
9. We view the order of the periodic table today according to the atomic number

Study Guide For Periodic Table

Jill Meryl Levy



Study Guide For Periodic Table:

Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03
The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section, and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts, and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step-by-step examples. Making it Real essays have been updated to present timely and engaging real-world applications, emphasizing the relevance of the material they are learning. This edition continues the end-of-chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter.

Hazmat Chemistry Study Guide (Second Edition) Jill Meryl Levy, 2011
[Periodic Table \(Advanced\)](#) Speedy Publishing, 2014-08-01
An advanced periodic table of elements displays not only the elements but the ions that form each element. A pamphlet with such a visual aid would greatly benefit chemistry students. Any student taking chemistry will need to learn the elements. A pamphlet would be concise and break the information down simply, making it easier to understand and remember. It allows students to simply focus on the main point rather than taking in information that they may or may not need.

Know Your 'O' Level Chemistry - A Study Guide , Safety Professional's Reference and Study Guide, Third Edition W. David Yates, 2020-03-19
This new edition serves both as a reference guide for the experienced professional and as a preparation source for those desiring certifications. It's an invaluable resource and a must-have addition to every safety professional's library. Safety Professional's Reference and Study Guide Third Edition is written to serve as a useful reference tool for the experienced practicing safety professional as well as a study guide for university students and those preparing for the Certified Safety Professional examination. It addresses major topics of the safety and health profession and includes the latest version of the Board of Certified Safety Professional (BCSP) reference sheet, a directory of resources and associations, as well as state and federal agency contact information. Additionally, this new edition offers new chapters and resources that will delight every reader. This book aids the prospective examination candidate and the practicing safety professional by showing them step-by-step how to solve each question formula listed on the BCSP examination and provide examples on how and when to utilize them.

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Peter L. Kresan, Reed Mencke, 2006-05-03
The guide helps students prepare for lectures and exams with a heavy emphasis on utilizing the book's Web resources.

[Barron's Science 360: A Complete Study Guide to Chemistry with Online Practice](#)
Barron's Educational Series, Mark Kernion, Joseph A. Mascetta, 2021-09-07
provides a complete guide to the fundamentals of

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Study Guide for The Anatomy and Physiology Learning System Edith Applegate, 2010-03-30 Designed to accompany The Anatomy and Physiology Learning System 4th Edition by Edith Applegate this study guide helps you learn and review basic A P concepts Each chapter emphasizes medical terminology with a set of key terms word parts clinical terms and abbreviations and then adds a variety of fun filled learning exercises review questions a quiz and a word puzzle The study guide corresponds to the textbook chapter for chapter Chapter learning objectives help you focus on the most important material Key concepts are defined on the first page of each chapter in the workbook Learning exercises for each chapter include short answer matching and diagrams to label and color Self quizzes allow you to measure your progress and understanding Fun and Games features end each chapter with a variety of engaging puzzles covering words and concepts A chapter summary provides a brief review of each chapter A chapter review provides questions for reinforcement and review of the concepts in each chapter

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