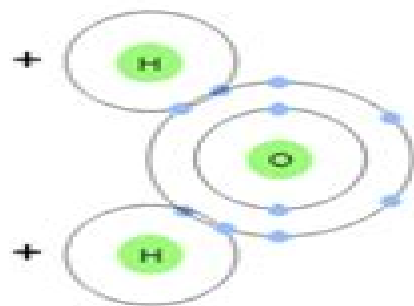


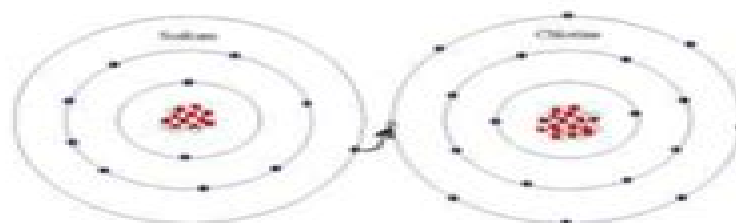
## Chemical Bonding (PS 4c)

Test date: \_\_\_\_\_

| Essential Concepts and Skills                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                      | Assignments                                           |
|------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|-------------------------------------------------------|
| <p><b>The Formation of Compounds through Ionic and Covalent Bonding (PS 4c)</b></p> <ul style="list-style-type: none"> <li>Metals tend to lose electrons in chemical reactions, forming positive ions. Nonmetals tend to gain electrons in chemical reactions, forming negative ions.</li> <li>Given a chemical formula of a compound, identify the elements and the number of atoms of each that comprise the compound.</li> <li>Recognize that the number of electrons on the outermost energy level (valence electrons) determines an element's chemical properties or chemical reactivity.</li> <li>Describe the difference between ionic and covalent bonding.</li> <li>Predict what kind of bond (ionic or covalent) will likely form when metals and nonmetals are chemically bonded.</li> <li>Gaining or losing electrons makes an atom an ion.</li> <li>Atoms react to form chemically stable substances that are held together by chemical bonds and are represented by chemical formulas. To become chemically stable, atoms gain, lose, or share electrons.</li> <li>Compounds are formed when elements chemically react. When a metallic element reacts with a nonmetallic element, their atoms gain and lose electrons respectively, forming ionic bonds. Generally, when two nonmetals react, atoms share electrons, forming covalent (molecular) bonds.</li> </ul> | <p>Pages 328-345<br/>Starter Quiz 1, 2<br/>Gizmos</p> |



**Covalent Bond**



**Ionic Bond**

# Study Guide Ionic And Covalent Bonding

**James E. Brady, Fred Senese**



## **Study Guide Ionic And Covalent Bonding:**

This book delves into Study Guide Ionic And Covalent Bonding. Study Guide Ionic And Covalent Bonding is a vital topic that needs to be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Study Guide Ionic And Covalent Bonding, encompassing both the fundamentals and more intricate discussions.

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- Chapter 1: Introduction to Study Guide Ionic And Covalent Bonding
- Chapter 2: Essential Elements of Study Guide Ionic And Covalent Bonding
- Chapter 3: Study Guide Ionic And Covalent Bonding in Everyday Life
- Chapter 4: Study Guide Ionic And Covalent Bonding in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Study Guide Ionic And Covalent Bonding. This chapter will explore what Study Guide Ionic And Covalent Bonding is, why Study Guide Ionic And Covalent Bonding is vital, and how to effectively learn about Study Guide Ionic And Covalent Bonding.
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4. In chapter 3, the author will examine the practical applications of Study Guide Ionic And Covalent Bonding in daily life. The third chapter will showcase real-world examples of how Study Guide Ionic And Covalent Bonding can be effectively utilized in everyday scenarios.
5. In chapter 4, this book will scrutinize the relevance of Study Guide Ionic And Covalent Bonding in specific contexts. This chapter will explore how Study Guide Ionic And Covalent Bonding is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, this book will draw a conclusion about Study Guide Ionic And Covalent Bonding. This chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Study Guide Ionic And Covalent Bonding.

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