

10.1 Periodic Trends



10.1 PERIODIC TRENDS FROM CONCEPTUAL MASTERY

Section 10.1 continued

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

_____	19
_____	20
_____	Aluminum
_____	14-18
_____	Transition?

Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 19. It exists in the periodic table.
- _____ 20. It has no line periodic table.
- _____ 21. Group 1 elements.
- _____ 22. Elements that are strong oxidizing agents.
- _____ 23. Group 18 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write one of the statements in terms of the statement in italics, change the underlined word or phrase to make it true.

- _____ 24. There are two representative elements of aluminum.
- _____ 25. Most elements have lines in the periodic table for metals.
- _____ 26. Group 14 elements except for hydrogen and boron in the alkali metals.
- _____ 27. Group 18 elements are the alkali metal metals.
- _____ 28. Group 14 elements are highly reactive transition elements in hydrogen.
- _____ 29. Group 14 elements are very reactive elements found in transition metals.
- _____ 30. Alkali metals have properties of both metals and non-metals.

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Edward J. Zalisko



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Table of Contents Study Guide For Content Mastery Chemistry Key

1. Understanding the eBook Study Guide For Content Mastery Chemistry Key
 - The Rise of Digital Reading Study Guide For Content Mastery Chemistry Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Study Guide For Content Mastery Chemistry Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Study Guide For Content Mastery Chemistry Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Study Guide For Content Mastery Chemistry Key
 - Personalized Recommendations
 - Study Guide For Content Mastery Chemistry Key User Reviews and Ratings
 - Study Guide For Content Mastery Chemistry Key and Bestseller Lists
5. Accessing Study Guide For Content Mastery Chemistry Key Free and Paid eBooks
 - Study Guide For Content Mastery Chemistry Key Public Domain eBooks
 - Study Guide For Content Mastery Chemistry Key eBook Subscription Services
 - Study Guide For Content Mastery Chemistry Key Budget-Friendly Options

6. Navigating Study Guide For Content Mastery Chemistry Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Study Guide For Content Mastery Chemistry Key Compatibility with Devices
 - Study Guide For Content Mastery Chemistry Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Study Guide For Content Mastery Chemistry Key
 - Highlighting and Note-Taking Study Guide For Content Mastery Chemistry Key
 - Interactive Elements Study Guide For Content Mastery Chemistry Key
8. Staying Engaged with Study Guide For Content Mastery Chemistry Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Study Guide For Content Mastery Chemistry Key
9. Balancing eBooks and Physical Books Study Guide For Content Mastery Chemistry Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Study Guide For Content Mastery Chemistry Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Study Guide For Content Mastery Chemistry Key
 - Setting Reading Goals Study Guide For Content Mastery Chemistry Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Study Guide For Content Mastery Chemistry Key
 - Fact-Checking eBook Content of Study Guide For Content Mastery Chemistry Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

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