

RETEACHING ACTIVITY 15

The Cultural Geography of Russia

Terms and Concepts

DIRECTIONS: Match each term from Chapter 15 with the correct definition.

- | | |
|------------------------------|--|
| <u>e</u> 1. czar | a. struggle between communism and capitalism |
| <u>f</u> 2. ethnic diversity | b. religious image |
| <u>b</u> 3. icon | c. philosophy based on Karl Marx's ideas |
| <u>d</u> 4. reform | d. change intended as improvement |
| <u>c</u> 5. communism | e. ruler of Russia before the revolution |
| <u>a</u> 6. Cold War | f. variety of ethnic groups |

Summarizing Information

DIRECTIONS: Read the passage below, and then answer the questions.

In 1921 the Bolsheviks, now known as Communists, won the civil war in Russia. The following year they established a new country, the Union of Soviet Socialist Republics (USSR), or the Soviet Union, with Moscow as the capital. Under the Communists, the Soviet Union gradually gained back Ukraine, Belorussia (now Belarus), much of the Caucasus region, and a large part of central Asia. After Lenin's death in 1924, Joseph Stalin, a leading Communist Party official, began a five-year climb to power. Defeating his rivals, Stalin set about making the Soviet Union into a powerful industrial giant.

7. What is the main topic of the passage?

Birth of the Soviet Union

8. What did the Communists accomplish after they won the civil war in Russia?

They established a new country, the USSR (Union of Soviet Socialist Republics) and gradually gained back territory.

9. In what new direction did Joseph Stalin take the country?

He set about making the Soviet Union into a powerful industrial giant.

(continued)

Us History Chapter 18 Section 3 Answer Key

Jennifer Gagliardi



Us History Chapter 18 Section 3 Answer Key:

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web other articles where acid base titration is discussed titration the sample and titrant include acid base titrations precipitation titrations complex formation titrations and oxidation reduction redox titrations in acid base titration i e the titration of an acid with a base or vice versa the indicator is a substance that can exist in two forms an

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web the above equation works only for neutralizations in which there is a 1 1 ratio between the acid and the base the example below demonstrates the technique to solve a titration problem for a titration of sulfuric acid with sodium hydroxide

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web there are many types of titration when considering goals and procedures however the most common types of titration in quantitative chemical analysis are redox titration and acid base titration titrations can be classified as acid base titrations redox titrations precipitation titrations complexometric titrations

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web nov 16 2023 acid base titration acid base titration is one of the four titration methods which is a quantitative method to analyze and determine an acids or bases concentration in the solution by neutralizing them precisely with the help of a

standard solution which could be of either base or acid of a known concentration

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web a titration experiment can be carried out to accurately measure the volume of substances that react in chemical reactions this technique is commonly used in neutralisation reactions and can also

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web titrant and analyte is a pair of acid and base acid base titrations are monitored by the change of pH as titration progresses let us be clear about some terminologies before we get into the discussion of titration curves titrant solution of a known concentration which is added to another solution whose concentration has to be determined

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web there are two basic types of acid base titrations indicator and potentiometric in an indicator based titration you add another chemical that changes color at the pH equal to the equivalence point when the acid and base are in stoichiometric proportions

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web acid base titrations are usually used to find the amount of a known acidic or basic substance through acid base reactions the analyte titrand is the solution with an unknown molarity the reagent titrant is the solution with a known molarity that will react with the analyte

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web as seen in the chapter on the stoichiometry of chemical reactions titrations can be used to quantitatively analyze solutions for their acid or base concentrations in this section we will explore the underlying chemical equilibria that make acid base titrimetry a

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volume v from 4 to 8 ml kg of ideal body weight ibw respiratory rate rr up to 35 bpm spo2 88 to 95 plateau pressure p less than 30 cm h2o ph goal 7.30 to 7.45 and inspiratory to expiratory time ratio less than 1

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