

SS 1 CHEMISTRY TEST QUESTIONS

SECTION A: OBJECTIVES

1. What is the primary basis for the arrangement of elements in the periodic table?
 - A) Atomic mass
 - B) Atomic number
 - C) Electron configuration
 - D) Physical properties
2. Which of the following elements belongs to the noble gas group?
 - A) Carbon
 - B) Oxygen
 - C) Neon
 - D) Nitrogen
3. How does atomic radius generally trend as you move from top to bottom across a period?
 - A) Increases
 - B) Decreases
 - C) Remains constant
 - D) No consistent trend
4. Which group of elements is characterized by high reactivity?
 - A) Alkali metals
 - B) Alkaline earth metals
 - C) Halogens
 - D) Noble gases
5. What is the term for the energy required to remove an electron from an atom?
 - A) Ionization energy
 - B) Electron affinity

Ss1 Test Question On Chemistry

B.R. Kowalski



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