

চট্টগ্রাম বোর্ড

১১। কোনটির বন্ধন শক্তি সর্বাধিক?

- ☒ (ক) নাইট্রোজেন
- (খ) অক্সিজেন
- (গ) হাইড্রোজেন
- (ঘ) ফ্লোরিন

১২। সাধারণ তাপমাত্রায় কোনটি তরল?

- (ক) C_2H_6
- (খ) C_4H_{10}
- ☒ (গ) C_6H_{14}
- (ঘ) $C_{20}H_{42}$

১৩। মাত্রিতে H^+ এর পরিমাণ বেড়ে গেলে কোনটি প্ররোপ করতে হবে?

- ☒ (ক) $CaCO_3$
- (খ) $CaCl_2$
- (গ) NH_4Cl
- (ঘ) $(NH_4)_2SO_4$

১৪। কোনটি উর্ধ্বপাতিত পদার্থ?

- (ক) ক্যালসিয়াম ক্লোরাইড
- ☒ (খ) অ্যামোনিয়াম ক্লোরাইড
- (গ) পটাশিয়াম ক্লোরাইড
- (ঘ) সোডিয়াম ক্লোরাইড

১৫। কোনটি উত্তেজক পদার্থ?

- (ক) বেনজিন
- (খ) মিথানল
- ☒ (গ) নাইট্রাস অক্সাইড
- (ঘ) নাইট্রো ট্রিসাইন

১৬। কোনটি অনুমোদিত দ্রুত বিজ্ঞানভিত্তিক?

- (ক) ক্যালসিয়াম কার্বাইড
- ☒ (খ) বেনজোয়িক এসিড
- (গ) অ্যাসিটিলিন
- (ঘ) ইথিলিন

১৭। ভারী ধাতু কোনটি?

- (ক) Fe
- (খ) Mo
- (গ) Rb
- ☒ (ঘ) Cd

১৮। PCl_3 মৌলের কেন্দ্রীয় পরমাণুর সর্বশেষ শক্তি স্তরে বন্ধন কোয় ইলেকট্রন সংখ্যা কতটি?

- ☒ (ক) 5
- (খ) 8
- (গ) 10
- (ঘ) 15

১৯। কোনটি বিজারক?

- (ক) Fe^{3+}
- (খ) Cu^{2+}
- (গ) Zn^{2+}
- ☒ (ঘ) Pb^{2+}

২০। C_3H_6 মৌলটি—

- I. সম্পৃক্ত হাইড্রোকার্বন
- II. পরিমাত্রকরণ বিক্রিয়া দেয়
- III. শক্তিশালী ও দুর্বল বন্ধন দ্বারা গঠিত

নিচের কোনটি সঠিক?

- (ক) I ও II
- (খ) I ও III
- ☒ (গ) II ও III
- (ঘ) I, II ও III

২১। সালফার ডাইঅক্সাইড মৌলে S এর সুস্থ যোজনী কত?

- (ক) 0
- ☒ (খ) 2
- (গ) 4
- (ঘ) 6

২২। 'X' একটি পলিমার বা পলিভিন এর তুলনায় কম নমনীয় এবং পানির পাইপ তৈরিতে ব্যবহৃত হয়।

'X' এর মনোমার কোনটি?

- (ক) $CH_2 = CH_2$
- ☒ (খ) $CH_2 = CHCl$
- (গ) $CH_2 = CH - CH_2$
- (ঘ) $CH = CH$

২৩। বেনজেনের বিজলপূরে কোনটি পাওয়া যায়?

- ☒ (ক) কেউলিন
- (খ) জিরকন
- (গ) বক্সাইট
- (ঘ) সিলভার

২৪। কোন মৌলটি অধিকতর তড়িৎ ঋণাত্মক?

- (ক) N
- (খ) O
- ☒ (গ) F
- (ঘ) Cl

২৫। কোনটির বাষ্পন হার কম?

- (ক) অ্যামোনিয়া
- ☒ (খ) নাইট্রোজেন
- (গ) ইথিলিন
- (ঘ) মিথন

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web each dipole represents one polar bond the molecule is nonpolar 10 describe what you think is happening to the penguins in the illustration of H_2O on page 4 they are being pulled towards the stronger polar bear he s more electronegative 11 explain what you think the crossed arrow represents in the comic book polar bonds perhaps 12

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web the two polar bears or penguins are equal in electronegativity therefore they pull electrons equally as hard and neither one will end up with more electrons more of the time 4 a what 3 types of bonds are represented on the third page of the comic book nonpolar covalent polar covalent and ionic b

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web lesson 16 polar bear and penguins this shows how the two atoms are bonded together with the two scoop of ice cream representing the bonded pairs of electrons the polar bear and penguin

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web answer because the two polar bears or penguins have equal electronegativity they will pull electrons equally hard and in most cases neither will end up with more electrons each polar bear and penguin has an electronegativity of the same value a

non polar bond is formed when two polar bears or two penguins are bonded together

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web why does the polar bear tell the penguin to try to be more positive the polar bear benefits more when the penguin is more positive the more the polar bear attracts the shared pair of electrons the more positive the partial charge on the penguin

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web answer the following questions consider the following illustration draw the lewis dot structure for hcl if the penguin represents a hydrogen atom and the polar bear represents a chlorine atom what does the ice cream represent in the drawing what do you think the picture is trying to illustrate would hcl be attracted to the charged wand

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web to answer this question you will examine 1 electronegativity 2 nonpolar vs polar molecules 3 electronegativity and bonding electronegativity the h atom and the cl atom in hydrogen chloride

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web 1 how does the comic strip define a polar molecule 2 define electronegativity as you understand it after reading the first two pages of the comic strip 3 what is the artist trying to represent by two polar bears arm wrestling or two penguins arm wrestling 4 what three types of bonds are represented on the third page of the comic strip

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web explain how the iceberg penguins and polar bears represent trends in electronegativity on the periodic table and more study with quizlet and memorize flashcards containing terms like how does the comic define a polar molecule define electronegativity as you understand it after reading the first two pages of the comic interpret the pic

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web so the final representation of the nh₃ molecule using polar bears and penguins is as follows begin array c text polar bear nitrogen text with two lone pairs of electrons text and three penguins hydrogen end array this is a fun and creative way to represent the nh₃ molecule

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web polar bears and penguins purpose in this lesson you will be exploring polarity and bonding between atoms in greater detail a comic book will provide new information about these topics and will introduce you to the concept of electronegativity which helps us understand partial charges