

- i) Suggest a suitable substance which could be used for the electrodes.

\_\_\_\_\_ 1mark

- ii) State the name of the gas given off at the electrode B?

\_\_\_\_\_ 1mark

- b) Write the reaction that takes place at each electrode when copper sulphate solution is electrolysed using :

- i) Graphite electrodes

- ii) Copper electrodes.

(4marks)

| Electrode          | Solution                 | Reaction takes place at the anode | Reaction takes place at the cathode. |
|--------------------|--------------------------|-----------------------------------|--------------------------------------|
| Graphite electrode | Copper sulphate solution |                                   |                                      |
| Copper electrode   | Copper sulphate solution |                                   |                                      |

- c) What mass of metallic silver will be deposited when a current of 0.53A is passed through a silver nitrate solution for 95 minutes? (mass Ag = 108, N = 14, O = 16)  
(4marks)

\_\_\_\_\_  
\_\_\_\_\_

# Sssscheme Of Work 4 Chemistry 2015 16

**JA Banks**



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