

STAAR Science Tutorial 07

TEK 6.5C: Elements & Compounds

TEK 6.5C: Differentiate between elements and compounds on the most basic level.

Elements & Atoms

- An **element** is a substance that cannot be separated into simpler substances by physical or chemical means. An element is already in its simplest form.
- The smallest piece of an element that still has the properties of that element is called an **atom**.
- An element is a pure substance, containing only one kind of atom. For example, the element gold is made only of gold atoms.
- Examples of elements include oxygen, potassium, carbon, hydrogen, sodium, and iron. The Periodic Table of Elements is a list of all the elements that have been discovered and named.
- Elements are represented on the Periodic Table by a one or two letter **symbol**. If it is a one letter symbol, it is always capitalized (such as S for sulfur). If it is a two letter symbol, the first letter is always capitalized and the second letter is always lowercase (such as He for helium).
- Most symbols look like abbreviations of the English element name. But some element symbols are based on the original Latin or Greek name for the element, such as gold (Au), silver (Ag), iron (Fe), sodium (Na), potassium (K) and mercury (Hg).

Compounds

- A **compound** is a substance that is made by chemically bonding two or more elements in a set ratio. For example, sodium (Na) and chlorine (Cl) in a 1:1 ratio makes the compound sodium chloride, commonly known as table salt. The symbols of the elements in a compound can be listed together to describe the chemical formula of the compound. Sodium chloride has the chemical formula "NaCl".
- Compounds have different chemical and physical properties than the elements from which they are made.
- For example, while sodium is a reactive metal and chlorine is a poisonous gas, the resulting white powder is neither a metal nor a gas, and is non-poisonous.

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