



LAGOS STATE GOVERNMENT
MINISTRY OF BASIC AND SECONDARY EDUCATION
3RD TERM UNIFIED EXAMINATION 2024

SS 1

CHEMISTRY

Time: 2 hrs.

SECTION A - OBJECTIVES

INSTRUCTION: Answer all questions in this section.

- The branch of science that deals with the nature and properties of substances and how one substance can be converted to another is known as
(a) Biology (b) Geography
(c) Chemistry (d) Physics
- The electron configuration $1s^2 2s^2 2p_x^2 2p_y^2$ depicts the
(a) Pauli's exclusion principle
(b) Aufbau's principle
(c) Octet rule
(d) Hund's rule
- The oxidation number of chromium in $Na_2Cr_2O_7$ is
(a) +2 (b) +6
(c) +7 (d) +12
- Bases normally
(a) are corrosive
(b) turn litmus paper from blue to red
(c) turn litmus paper from red to blue
(d) are non-metal oxides
- Naturally occurring Boron is made up of 19.9% ^{10}B and 80.1% ^{11}B . The relative atomic mass of Boron is
(a) 21.0 (b) 10.8
(c) 10.5 (d) 10.0
$$\frac{19.9 \times 10 + 80.1 \times 11}{100} = 10.81$$
- Isotopes of the same element have similar chemical properties because they have the same number of
(a) Nuclides (b) Neutrons
(c) Atoms (d) protons
- Which of the following household liquids would be effective in treating someone with too much acid in the stomach?
(a) Strong salt solution with pH 7.0
(b) Tomato juice with pH 4.1
(c) Milk of Magnesia with pH 10.5
(d) Black coffee with pH 5.0
- The percentage by mass of oxygen in $MgSO_4 \cdot 7H_2O$ is
(a) 26.0% (b) 45.5%
(c) 71.5% (d) 84.0%
- Ethane of volume $400cm^3$ was completely burnt in excess oxygen according to the following equation:
 $2C_2H_6(g) + 7O_2(g) \rightarrow 6H_2O(g) + 4CO_2(g)$
Calculate the volume of steam that would be produced.
(a) $200cm^3$ (b) $400cm^3$
(c) $600cm^3$ (d) $1200cm^3$
$$\frac{2 \times 400}{2} = 400 \quad \frac{6 \times 1200}{4} = 1800$$

$$400 = x \quad 1800 = 1200$$
- A weak acid is one which
(a) is not corrosive
(b) completely ionizes in water
(c) does not produce salt with alkali
(d) slightly ionizes in water

Sss1 Chemistry Quiz

J Dewey



Sss1 Chemistry Quiz:

Environmental Chemistry (Volume-I) Dr. Benvikram Barman, Dr. Bhupendra Singh Banjare, Dr. Deepak Kumar Sahu, Dr. Swati Chandrawanshi , Dr. Shobhana Ramteke, Ms. Rashmita Khuntia, 2025-09-19 Environmental Chemistry Volume I DSE Based on Syllabus of NEP 2020 is a comprehensive textbook that explores the intricate relationship between chemistry and the environment Designed for students educators and researchers this volume presents a planned study of fundamental environmental concepts and processes integrating both theoretical and applied aspects The book begins with a strong foundation in environmental chemistry and systematically covers the essential components of the environment including the biosphere hydrosphere lithosphere and atmosphere It emphasizes ecological principles classification of ecosystems and the functioning of biogeochemical cycles highlighting the circulation of nitrogen carbon phosphorus hydrogen and oxygen which are vital for sustaining life on Earth Further the volume addresses critical environmental issues such as thermal pollution noise pollution and water related concerns Detailed discussions include the origin physico chemical properties and sources of water the hydrological cycle criteria of water quality and modern water management practices like watershed management and rainwater harvesting The book also provides insights into water pollution its sources harmful effects and control strategies The atmospheric portion of the text examines the structure major regions and composition of the atmosphere along with an in depth discussion on temperature inversion and its role in air pollution episodes This volume serves as an essential resource for undergraduate and undergraduate postgraduate students of chemistry environmental science and related fields It will also benefit academicians and researchers with sustainable environmental management *Annual report of the State Board of Health of Illinois. 1894 ,1894* *Chemistry Quiz Book* Mrinal Talukdar, 2021-01-19 Dive into the fascinating world of chemistry with Chemistry Quiz Book by Mrinal Talukdar an engaging and informative exploration of the fundamental principles concepts and applications of this essential science Whether you re a student educator or simply curious about the wonders of chemistry this quiz book offers a fun and interactive way to test your knowledge and deepen your understanding of the subject In this comprehensive quiz book Mrinal Talukdar presents a wide range of questions covering various topics in chemistry from the periodic table and chemical bonding to organic chemistry and chemical reactions With quizzes designed to challenge and educate readers of all levels this book provides an enjoyable way to learn about the building blocks of matter the properties of elements and the processes that govern chemical reactions Through engaging quizzes and thought provoking questions readers have the opportunity to test their understanding of key concepts and principles in chemistry Whether you re exploring the basics of atomic structure or delving into advanced topics like thermodynamics and kinetics this quiz book offers a fun and interactive way to reinforce your knowledge and enhance your learning experience The overall tone and mood of Chemistry Quiz Book are one of curiosity and discovery as Talukdar invites readers to explore the fascinating world of chemistry through a series of entertaining quizzes and exercises With its

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A Whole Year of Chemistry Quizzes Siegfried Howell, 2019-12-12 This book *A Whole Year of Chemistry Quizzes* was written to provide easy to use and grade quizzes to assess the comprehension of honors students Advance Placement students AP and International Baccalaureate IB students This book of quizzes guides the teacher and the student through what is required in a non watered down chemistry course that leads students towards test and college readiness The outline of this book has a minimum of 4 quizzes per chapter that prepares students for the formative assessment associated at the end of all chapters The 25 chapters include topics that are covered in the honors chemistry setting as well as specialty topics like thermodynamics kinetics rates of reactions that are seen in the Advance Placement classes Included within this book are quizzes for the International Baccalaureate teacher that wishes to test students on environmental chemistry as well as biological and food chemistry This is a book that was written to fill the void of valuable resources needed for novice and experienced teachers in institutions that continually push for more summative assessments higher DOKs and rapid feedback while limiting preparation time As a teacher for over 25 years I know that any well outlined structured and comprehensive resource saves time in additional planning searching and preparing Use this book to help you identify and test students on topics that are important to their comprehension and success with their final test Chapter 1 Matter and change Chapter 2 measurement and calculations Chapter 3 Atoms The building blocks of matter Chapter 4 Arrangement of electrons in atoms Chapter 5 The periodic law Chapter 6 Chemical bonding Chapter 7 Chemical formulas and chemical compounds Chapter 8 Chemical equations and reactions Chapter 9 Stoichiometry Chapter 10 Physical characteristics of gases Chapter 11 Molecular composition of gases Chapter 12 Liquids and solids Chapter 13 Solutions Chapter 14 Ions in aqueous solution and colligative properties Chapter 15 Acids and bases Chapter 16 Acid base titrations Chapter 17 Reaction energy and reaction kinetics Chapter 18 Chemical equilibrium Chapter 19 Oxidation reduction reactions Chapter 20 Chemical thermodynamics Chapter 21 Carbon and hydrocarbons Chapter 22 Other organic compounds Chapter 23 Nuclear chemistry Chapter 24 Biological and Food chemistry Chapter 25 Environmental chemistry

General Chemistry Quiz Book, Problems and Questions Walter H. Beisler, 1926

College Chemistry Quiz Book Cecil Victor King, Worth Wade, James Kendall, 1930

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