

- ____ 1.5. The simplest carboxylic acid is commonly known as ____.
- acetic acid
 - acetic
 - formic acid
 - formic
- ____ 1.6. Which of the following tend to be basic?
- amide
 - amine
 - alcohol
 - ether
- ____ 1.7. Which of the following is a type of addition reaction?
- hydrogenation
 - dehydrogenation
 - halogenation
 - hydrohalogenation

Matching

Match the terms below with their correct definitions.

- | | |
|------------------------|--------------------------------|
| a. ether | h. polymer |
| b. ketone | i. carbonyl group |
| c. functional group | m. monomer |
| d. isomers and alcohol | n. addition polymerization |
| e. amide | o. carbonyl group |
| f. halomethane | p. ester |
| g. alkylhalide | q. condensation polymerization |
| h. hydroxyl group | r. HPC |
| i. carboxylic acid | s. hydration reaction |
| j. amine | |

- ____ 18. In an organic molecule, any atom or group of atoms that acts as a substituent
- ____ 19. Large molecule consisting of many repeating subunits
- ____ 20. Ethanol with small amounts of monomer materials added
- ____ 21. Compound containing a nitrogen atom bonded to a carbon atom in an aliphatic chain or aromatic ring
- ____ 22. Compound with a carbonyl group bonded to two carbon atoms
- ____ 23. Compound in which the acidic hydrogen of a carboxyl group is replaced by a carbon atom
- ____ 24. A group in which a carbonyl group is bonded to a hydroxyl group
- ____ 25. Compound with an oxygen atom bonded to two carbon atoms
- ____ 26. Hydrogen and fluorine atoms bonded to carbon to produce a hydrofluorocarbon
- ____ 27. Compound containing a carbonyl group bonded to a carbon atom and to a hydrogen atom
- ____ 28. Reaction in which all atoms in the monomer are in the polymer
- ____ 29. Reaction in which monomers with at least two functional groups combine to form a polymer, with the loss of a small by-product
- ____ 30. Compound in which the -OH group of a carboxylic acid is replaced by a nitrogen atom bonded to other atoms

Substituted Hydrocarbons And Their Reactions Answers

**Dr Vijay Sarda, Dr A C Handa, Dr K K
Arora**



Substituted Hydrocarbons And Their Reactions Answers:

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Frontier Orbitals and Reaction Paths Kenichi Fukui, Hiroshi Fujimoto, 1997 This book is a collection of selected papers on the Frontier Orbital Theory by Nobel prizewinner Kenichi Fukui Chemistry 1981 with introductory notes It provides the basic concept and formulation of the theory and the physical and chemical significance of the frontier orbital interactions in chemistry together with many practical applications The formulation of the Intrinsic Reaction Coordinate and applications to some simple systems are also presented The aim of this volume is to show by what forces chemical reactions are driven and to demonstrate how the regio and stereo selectivities are determined in chemical reactions Students and senior investigators will gain insight into the nature of chemical reactions and find out how quantum chemical calculations are connected with chemical intuition

DAT: Dental Admissions Test: Includes 3 Full Length Practice Tests + Online Access to Video Tutorials Barron's Educational Series, Joseph DiRienzo, John J. Ference, Nicole D. Cornell, Edwin H. Hines, John Swartwood, 2018-05-15 This brand new manual prepares dental school applicants across the United States and Canada to pass the required admissions test It features Three full length model tests including a diagnostic test All answers explained in detail Access to video tutorials from the authors and more Test takers will also find thorough reviews of all DAT test topics a general survey of the natural sciences including biology chemistry and organic chemistry as well as testing for perceptual ability reading comprehension and quantitative reasoning ONLINE PRACTICE TEST Students will also get access to one additional full length online DAT test with all questions answered and explained This online exam can be easily accessed by smartphone tablet or computer

Science Assertion and Reason Class 10 Priti Singhal, 2024-11-14 This book is structured to align with the latest syllabus and curriculum guidelines ensuring the content is relevant and rigorous Each chapter begins with a clear set of learning objectives providing a roadmap for students to understand what they will achieve by the end of the chapter We have included numerous diagrams illustrations and real life examples to make complex concepts more accessible and engaging

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