

Copper In Silver Nitrate Lab

by Charles Deaton

Background

The purpose of this lab is to find the theoretical percentage yield of the reaction. The reaction consisted of copper(Cu) and silver nitrate(AgNO_3), which after 3 class days of reacting, turned into copper nitrate($\text{Cu}(\text{NO}_3)_2$), and silver(Ag). In that third class day, they massed that 3 inch copper wire, and 20 ml of silver nitrate solution were measured. Once it was measured, the wire hung over the side of a small beaker with a good portion of the wire submerged in 20 ml of solution. Observations were made and the three beakers were left out until day 3. When students returned a couple days later, and measured what they found to be a beaker. Stripping their beaker out of the solution and submerging with water so that the precipitate were over the beaker. This mass of the wire was taken again, and a piece of filter paper was also weighed. The filter paper was then set up in a ring stand with a funnel and a large beaker beneath. The concentrated blue beaker water passed through the filter paper, and mixed with more water. This was for removing all of the reactants from the beaker. Day 4 was to remove the final mass of our products so that filter paper that could be dry from the days prior. After everything was measured and written in the table, calculations were done for finding theoretical yield, percent yield, and percent error of the lab.

Silver Nitrate Lab Report

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Silver Nitrate Lab Report:

Chemistry in the Laboratory James M. Postma, Julian L. Robert, J. Leland Hollenberg, 2004-03-12 This clearly written class tested manual has long given students hands on experience covering all the essential topics in general chemistry Stand alone experiments provide all the background introduction necessary to work with any general chemistry text This revised edition offers new experiments and expanded information on applications to real world situations **Annual reports of the Chemical Laboratory of the American Medical Association. v. 7, 1914** ,1914 Water-resources Investigations Report ,1998 *Breathe of the Flesh* Jack Allen, 2002-06 It s 1942 the middle of WWII New York City is filthy with German spies FBI agent Thomas Leopard is drinking and suicidal selfish loathsome and hateful And he has a killer loose in his city the German spy Der Tiger who has been dormant up to that point of the war Now the Abwehr the intelligence branch of the Nazi military has a special mission for its most lethal and dangerous spy Annual Report of the North Dakota Agricultural Experiment Station North Dakota Agricultural Experiment Station (Fargo), 1908 **Publications, Reports and Papers for ... from Oak Ridge National Laboratory** Oak Ridge National Laboratory, 1964 **Report of the Lake Michigan Water Commission** Lake Michigan Water Commission, 1909 Scientific and Technical Aerospace Reports ,1987 *Bulletin of the Bureau of Standards* United States. Bureau of Standards, 1917 Contains results of investigations researches etc pertaining to scientific technical and manufacturing interests of the country **Subject Index to Unclassified ASTIA Documents** Defense Documentation Center (U.S.), 1960 *Keywords Index to U.S. Government Technical Reports* United States. Department of Commerce. Office of Technical Services, 1963 **Laboratory Methods in Dynamic Electroanalysis** M. Teresa Fernández Abedul, 2019-10-13 Laboratory Methods in Dynamic Electroanalysis is a useful guide to introduce analytical chemists and scientists of related disciplines to the world of dynamic electroanalysis using simple and low cost methods The trend toward decentralization of analysis has made this fascinating field one of the fastest growing branches of analytical chemistry As electroanalytical devices have moved from conventional electrochemical cells 10 20 mL to current cells e g 5 50 mL based on different materials such as paper or polymers that integrate thick or thin film electrodes interesting strategies have emerged such as the combination of microfluidic cells and biosensing or nanostructuration of electrodes This book provides detailed easy procedures for dynamic electroanalysis and covers the main trends in electrochemical cells and electrodes including microfluidic electrodes electrochemical detection in microchip electrophoresis nanostructuration of electrodes development of bio enzymatic immuno and DNA assays paper based electrodes interdigitated array electrodes multiplexed analysis and combination with optics Different strategies and techniques amperometric voltammetric and impedimetric are presented in a didactic practice based way and a bibliography provides readers with additional sources of information Provides easy to implement experiments using low cost simple equipment Includes laboratory methodologies that utilize both conventional designs and the latest trends in dynamic electroanalysis Goes beyond

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