	7 1
SCHOOL.	DATE
7	SALTS
198	89 Q 34b,c
	(a) Explain how sodium carbonate can be used to soften hard water. (Use jonic equations where necessary). (3 marks)
	(Use ionic equations where necessary). (3 marks)
	(b) Give one other commercial use of sodium carbonate, besides softening of har
	water. (1 mark
1000	PPIA OUESTION 7
1303	What would be observed when aqueous sodium hydroxide is added to aqueous lead (II) nitrate?
	PPIA QUESTION 18
P. 6	PPIA QUESTION 18 Q and R are solutions of lead nitrate, sodium chloride and magnesium
P. 6	PPIA QUESTION 18 Q and R are solutions of lead nitrate, sodium chloride and magnesium lphate (not in that order) Use the following information to identify P and Q When drops of barium chloride solutions are added to separate portions of P. (
P. C Sul	PPIA QUESTION 18 Q and R are solutions of lead nitrate, sodium chloride and magnesium lphate (not in that order) Use the following information to identify P and Q When drops of barium chloride solutions are added to separate portions of P, Q and R, a white P is formed
P. (Sui (i)	PPIA QUESTION 18 Q and R are solutions of lead nitrate, sodium chloride and magnesium lphate (not in that order) Use the following information to identify P and Q When drops of barium chloride solutions are added to separate portions of P, Q and R, a white P is formed
P. (Sui (i)	PPIA QUESTION 18 Q and R are solutions of lead nitrate, sodium chloride and magnesium liphate (not in that order) Use the following information to identify P and Q. When drops of barium chloride solutions are added to separate portions of P, Q and R, a white P is formed. When dilute hydrochloric acid is added to separate portions of Q and R, a white

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W. Terry Whalin, Terry Whalin

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