



Equations

Name _____

Date _____

Write a balanced chemical equation to represent each of the following chemical reactions.

1. iron + sulfur $\rightarrow$ iron(II) sulfide
2. zinc + copper(II) sulfate $\rightarrow$ zinc sulfate + copper
3. silver nitrate + sodium bromide $\rightarrow$ sodium nitrate + silver bromide
4. potassium chlorate (heated) $\rightarrow$ potassium chloride + oxygen
5. water (electricity) $\rightarrow$ hydrogen + oxygen
6. mercury(II) oxide (heated) $\rightarrow$ mercury + oxygen
7. potassium iodide + lead(II) nitrate $\rightarrow$ lead(II) iodide + potassium nitrate
8. aluminum + oxygen $\rightarrow$ aluminum oxide
9. magnesium chloride + ammonium nitrate $\rightarrow$ magnesium nitrate + ammonium chloride
10. iron(III) chloride + ammonium hydroxide $\rightarrow$ iron(III) hydroxide + ammonium chloride
11. sodium peroxide + water $\rightarrow$ sodium hydroxide + oxygen
12. iron(III) oxide + carbon $\rightarrow$ iron + carbon monoxide
13. iron + water $\rightarrow$ hydrogen + iron(III) oxide
14. iron(III) chloride + potassium hydroxide $\rightarrow$ potassium chloride + iron(III) hydroxide
15. aluminum + sulfuric acid $\rightarrow$ aluminum sulfate + hydrogen
16. sodium carbonate + calcium hydroxide $\rightarrow$ sodium hydroxide + calcium carbonate



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Julia Schneider



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Table of Contents Weston Walch Publisher Spanish Activities Answers

1. Understanding the eBook Weston Walch Publisher Spanish Activities Answers
 - The Rise of Digital Reading Weston Walch Publisher Spanish Activities Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Weston Walch Publisher Spanish Activities Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Weston Walch Publisher Spanish Activities Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Weston Walch Publisher Spanish Activities Answers

- Personalized Recommendations
 - Weston Walch Publisher Spanish Activities Answers User Reviews and Ratings
 - Weston Walch Publisher Spanish Activities Answers and Bestseller Lists
5. Accessing Weston Walch Publisher Spanish Activities Answers Free and Paid eBooks
- Weston Walch Publisher Spanish Activities Answers Public Domain eBooks
 - Weston Walch Publisher Spanish Activities Answers eBook Subscription Services
 - Weston Walch Publisher Spanish Activities Answers Budget-Friendly Options
6. Navigating Weston Walch Publisher Spanish Activities Answers eBook Formats
- ePub, PDF, MOBI, and More
 - Weston Walch Publisher Spanish Activities Answers Compatibility with Devices
 - Weston Walch Publisher Spanish Activities Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Weston Walch Publisher Spanish Activities Answers
 - Highlighting and Note-Taking Weston Walch Publisher Spanish Activities Answers
 - Interactive Elements Weston Walch Publisher Spanish Activities Answers
8. Staying Engaged with Weston Walch Publisher Spanish Activities Answers
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Weston Walch Publisher Spanish Activities Answers
9. Balancing eBooks and Physical Books Weston Walch Publisher Spanish Activities Answers
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Weston Walch Publisher Spanish Activities Answers
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Weston Walch Publisher Spanish Activities Answers
- Setting Reading Goals Weston Walch Publisher Spanish Activities Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Weston Walch Publisher Spanish Activities Answers

- Fact-Checking eBook Content of Weston Walch Publisher Spanish Activities Answers
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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