

Chemistry

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(Chapter 12) (Aldehydes, Ketones and Carboxylic Acids) (Intext Questions)
(Class 12)

Question 12.4:

Arrange the following compounds in increasing order of their reactivity in nucleophilic addition reactions.

(i) Ethanal, Propanal, Propanone, Butanone.

(ii) Benzaldehyde, *p*-Tolualdehyde, *p*-Nitrobenzaldehyde, Acetophenone.

Hint: Consider steric effect and electronic effect.

Answer 12.4:

(i) The +I effect of the alkyl group increases in the order:

Ethanal < Propanal < Propanone < Butanone

The electron density at the carbonyl carbon increases with the increase in the +I effect. As a result, the chances of attack by a nucleophile decrease. Hence, the increasing order of the reactivities of the given carbonyl compounds in nucleophilic addition reactions is:

Butanone < Propanone < Propanal < Ethanal



Ethanal



Propanal



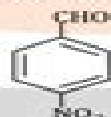
Propanone



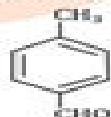
Butanone

(ii) The +I effect is more in ketone than in aldehyde. Hence, acetophenone is the least reactive in nucleophilic addition reactions. Among aldehydes, the +I effect is the highest in *p*-tolualdehyde because of the presence of the electron-donating -CH₃ group and the lowest in *p*-nitrobenzaldehyde because of the presence of the electron-withdrawing -NO₂ group. Hence, the increasing order of the reactivities of the given compounds is:

Acetophenone < *p*-tolualdehyde < Benzaldehyde < *p*-Nitrobenzaldehyde



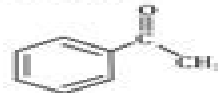
p-Nitrobenzaldehyde



p-Tolualdehyde



Benzaldehyde



Acetophenone

Question 12.5:

Predict the products of the following reactions:

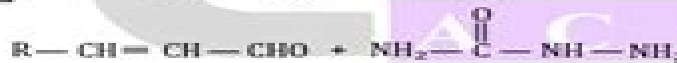
(i)



(ii)



(iii)



(iv)



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Table of Contents Solved Exercises Of 12th Chemistry

1. Understanding the eBook Solved Exercises Of 12th Chemistry
 - The Rise of Digital Reading Solved Exercises Of 12th Chemistry
 - Advantages of eBooks Over Traditional Books
2. Identifying Solved Exercises Of 12th Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
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 - Popular eBook Platforms
 - Features to Look for in an Solved Exercises Of 12th Chemistry
 - User-Friendly Interface
4. Exploring eBook Recommendations from Solved Exercises Of 12th Chemistry

- Personalized Recommendations
- Solved Exercises Of 12th Chemistry User Reviews and Ratings
- Solved Exercises Of 12th Chemistry and Bestseller Lists
- 5. Accessing Solved Exercises Of 12th Chemistry Free and Paid eBooks
 - Solved Exercises Of 12th Chemistry Public Domain eBooks
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 - Solved Exercises Of 12th Chemistry Budget-Friendly Options
- 6. Navigating Solved Exercises Of 12th Chemistry eBook Formats
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 - Solved Exercises Of 12th Chemistry Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Solved Exercises Of 12th Chemistry
 - Highlighting and Note-Taking Solved Exercises Of 12th Chemistry
 - Interactive Elements Solved Exercises Of 12th Chemistry
- 8. Staying Engaged with Solved Exercises Of 12th Chemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Solved Exercises Of 12th Chemistry
- 9. Balancing eBooks and Physical Books Solved Exercises Of 12th Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Solved Exercises Of 12th Chemistry
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Solved Exercises Of 12th Chemistry
 - Setting Reading Goals Solved Exercises Of 12th Chemistry
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Solved Exercises Of 12th Chemistry

- Fact-Checking eBook Content of Solved Exercises Of 12th Chemistry
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
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