

DSLR-A100

SERVICE MANUAL

Ver 1.3 2006.10

Revision History

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DISASSEMBLY	SCHEMATIC DIAGRAMS	

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Sony Alpha Dslr A100 Service Manual Level 1 2 3 Adjustments Manual

Wenbin Ji



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web oct 2 2019 let s start by balancing the oxygen atoms to do this make the oxygen atoms as six on either side of the chemical equation $2\text{Fe}_2\text{O}_3 \rightarrow 4\text{Fe} + 3\text{CO}_2$ on towards the next step now step 3 once you are done with this check out if the number of atoms for the other elements is also equal on both the sides

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there are equal numbers of atoms of each element on both sides of the equation we can write a chemical equation for the reaction of carbon with hydrogen gas to form methane CH_4

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$$\begin{array}{l} 1 \text{ N}_2 + 3 \text{ H}_2 \rightarrow 2 \text{ NH}_3 \\ 3 \text{ KClO}_3 \rightarrow 3 \text{ KCl} + 3 \text{ O}_2 \\ 2 \text{ NaCl} + 1 \text{ F}_2 \rightarrow 2 \text{ NaF} \\ 1 \text{ Cl}_2 + 4 \text{ H}_2 \rightarrow 2 \text{ H}_2\text{O} + 2 \text{ H}_2\text{O} \\ 5 \text{ Pb} + 2 \text{ OH}^- + 2 \text{ HCl} \rightarrow 2 \text{ H}_2\text{O} + 1 \text{ PbCl}_2 \\ 2 \text{ Al} + 3 \text{ Br}_2 \rightarrow 2 \text{ AlBr}_3 \\ 3 \text{ K}_2\text{SO}_4 + 6 \text{ KBr} \rightarrow 1 \text{ Al}_2\text{SO}_4 + 3 \text{ H}_2\text{O} \\ 1 \text{ C}_3\text{H}_8 + 5 \text{ O}_2 \rightarrow 3 \text{ CO}_2 + 4 \text{ H}_2\text{O} \end{array}$$

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web may 28 2020 write a balanced chemical equation for each step of the process the first step is the decomposition of solid calcium carbonate from seashells to form solid calcium oxide and gaseous carbon dioxide the second step is the formation of solid calcium hydroxide as the only product from the reaction of the solid calcium oxide with liquid water