

When iron(II) sulfate is used for killing weeds in lawns, it is often mixed with the fertiliser ammonium sulfate. Ammonium sulfate also makes the soil acidic.

- (a) Write an equation to show how the ammonium ion behaves as a Brønsted–Lowry acid in water.

.....

(1)

- (b) Compounds such as ammonium sulfate react on warming with sodium hydroxide solution as shown in the equation below.



Use this information to describe a simple test, other than smell, to show that ammonia is evolved. State what you would observe.

Test

.....

Observation

(2)

(Total 3 marks)

Titration Questions Gcse

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