

## Mole Concept

### Mole concept:

Mole is the counting unit for atoms, molecules and ions and is equal to  $6.022 \times 10^{23}$ . This number is called Avogadro's number and is denoted by  $N_A$ .

One mole is the amount of substance that contains as many particles (atoms, molecules or ions) as in 12g of C-12.

### Molar mass:

It is the mass of 1 mole of any substance expressed in grams.

In case of atoms it is equal to atomic mass expressed in grams.

In case of molecules it is equal to molecular mass expressed in grams.

1 Mole of  
particles  
(atoms,  
molecules or  
ions)

Contains  $6.022 \times 10^{23}$  particles  
(atoms,  
molecules or  
ions)

Mass is  
equal to  
molar mass  
(Atomic mass  
or molecular  
mass)

Volume is  
equal to  
molar  
volume =  
22.4 lit

For example,

1 mole of Oxygen gas	Contains $6.022 \times 10^{23}$ Oxygen molecules.	32 g	22.4 lit
1 mole of Oxygen atoms	Contains $6.022 \times 10^{23}$ Oxygen atoms.	16g	22.4 lit
1 mole of Carbon dioxide gas	Contains $6.022 \times 10^{23}$ Carbon dioxide molecules.	44g	22.4 lit
1 mole of Ammonia gas	Contains $6.022 \times 10^{23}$ Ammonia molecules.	17g	22.4 lit

### Formula

1. No. of moles = given mass / molar mass ( $n = m/M$ )
2. No. of moles = given no. of particles / Avogadro no. ( $n = N/N_A$ )
3. Molar mass = given mass / no. of moles ( $M = m/n$ )
4. Mass of the substance = molar mass  $\times$  no. of moles ( $m = M \times n$ )
5. No. of particles = no. of moles  $\times$  Avogadro's no. (no. of particles =  $n \times N_A$ )  
= (Given Mass/ Molar mass)  $\times N_A$

# The Mole Concept Review Key

**James E. Brady, Fred Senese**



## **The Mole Concept Review Key:**

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*Daily Graphic* Elvis Aryeh, 2002-05-07      *SAT Subject Test: Chemistry with Online Tests* Joseph A. Mascetta, Mark Kernion, 2018-09-01. Always study with the most up to date prep. Look for SAT Subject Test Chemistry ISBN 9781506263120 on sale December 01 2020. Publisher's Note: Products purchased from third party sellers are not guaranteed by the publisher for quality, authenticity, or access to any online entitlements included with the product.

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