

CHAPTER 5 REVIEW*The Periodic Law***MIXED REVIEW****SHORT ANSWER** Answer the following questions in the space provided.

1. Consider a neutral atom with 53 protons and 74 neutrons to answer the following questions.
 - _____ a. What is its atomic number?
 - _____ b. What is its mass in amu's?
 - _____ c. Is the element's position in a modern periodic table determined by its atomic number or by its atomic mass?
2. Consider an element whose outermost electron configuration is $3d^{10}4s^24p^x$.
 - _____ a. To which period does the element belong?
 - _____ b. If it is a halogen, what is the value of x ?
 - _____ c. The group number will equal $(10 + 2 + x)$. True or False?
3. _____ a. Metalloids are found in which block, s , p , d , or f ?
- _____ b. The hardest, densest metals are found in which block, s , p , d , or f ?
4. _____ a. Name the most chemically active halogen.
- _____ b. Write its electron configuration.
- _____ c. Write the configuration of the most-stable ion this element makes.
5. Referring only to the periodic table at the top of the Section 5-2 Review on page 35, answer the following questions on periodic trends.
 - _____ a. Which has the larger radius, Al or In?
 - _____ b. Which has the larger radius, Se or Ca?
 - _____ c. Which has a larger radius, Ca or Ca^{2+} ?
 - _____ d. Which has greater ionization energies as a class, metals or nonmetals?
 - _____ e. Which has the greater ionization energy, As or Cl?
 - _____ f. An element with a large negative electron affinity is most likely to form a positive ion, a negative ion, or a neutral atom?

The Periodic Law Mixed Review Worksheet

**Holt Rinehart & Winston, Holt,
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Physical Review, 1927 Vols for 1903 include Proceedings of the American Physical Society

The Boston Journal of Chemistry and Popular Science Review, 1881

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Chemical News and Journal of Physical Science William Crookes, James H. Gardiner, Gerald Druce, H. W. Blood-Ryan, 1913

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