

Question Number	Acceptable Answers	Reject	Mark
14(a)	So that the phenol is used up / methyl orange is bleached before the rate changes (significantly) OR So that the phenol is used up / methyl orange is bleached during the initial rate period OR So that the concentration of bromide/bromate/reactants does not fall significantly before all the phenol is used up / the methyl orange is bleached OR Within this region/period/time the average rate of reaction approximates to the initial rate	bromine	(1)

Question Number	Acceptable Answers	Reject	Mark
14(b)(i)	So that only the concentration of bromide ions varies (significantly) during the course of the reaction / so that the concentration of the bromide ions is the limiting factor / so that the concentration of bromide ions is the only variable ALLOW So their concentrations / the BrO_3^- and H^+ concentrations do not change OR So their concentrations / the BrO_3^- and H^+ concentrations are not the limiting factor		(1)

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 $HCl(aq) + NaOH(aq) \rightarrow NaCl(aq) + H_2O(l)$
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