

Lecture 3: Part 2

Polyatomic Ions, Ionic Reactions and Redox Reactions

- Polyatomic ions are covalent molecules that possess charge and behave as normal ions in solution.
- When a salt containing a polyatomic ion is dissolved in water, the polyatomic ions themselves DO NOT break apart. They are simply separated from the *counter-ion*.
- Example: *Phosphate* (PO_4^{3-})



Sodium
Phosphate

Sodium
cations

Phosphate
anion

What Is A Polyatomic Ion Packets Chemical Reactions

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