

Periodic Table Named-Groups Worksheet – 2

DIRECTIONS: Use the Periodic Table shown below to answer each question. using the letters in each "box" to identify the required element(s).

The diagram shows a crossword puzzle grid with the following letters placed in specific cells:

- A**: Top left corner.
- B**: Top right corner.
- C**: Second row from top, second column from left.
- D**: Second row from top, eighth column from left.
- E**: Second row from top, tenth column from left.
- F**: Third row from top, second column from left.
- G**: Third row from top, sixth column from left.
- H**: Third row from top, eighth column from left.
- I**: Fourth row from top, tenth column from left.
- J**: Fifth row from top, second column from left.
- K**: Fifth row from top, twelfth column from left.
- L**: Fifth row from top, third column from left.
- M**: Fifth row from top, sixth column from left.
- N**: Fifth row from top, eighth column from left.
- O**: Fifth row from top, tenth column from left.
- P**: Fifth row from top, twelfth column from left.
- Q**: Sixth row from top, second column from left.
- R**: Sixth row from top, third column from left.
- S**: Sixth row from top, sixth column from left.
- T**: Sixth row from top, seventh column from left.
- U**: Sixth row from top, eighth column from left.
- V**: Seventh row from top, second column from left.
- W**: Seventh row from top, eighth column from left.
- X**: Eighth row from top, second column from left.
- Y**: Eighth row from top, eighth column from left.
- Z**: Eighth row from top, sixth column from left.

Arrows point from the letters **R** and **Z** to a separate section below the grid, which contains the letters **V**, **W**, **X**, and **Y** in a similar grid layout.

List elements in alphabetical order when more than 1 element answers a question.

- _____ 1) Element **A** is in a box all by itself because: **A)** it is really a Noble Gas **B)** it is a gas **C)** it is NOT a gas **D)** it does NOT share the properties of the other elements in its group **E)** it is a liquid
- _____ 2) Which of these elements is/are highly reactive? **A)** "C" **B)** "E" **C)** "K" **D)** A, B, and C **E)** A and B only
- _____ 3) What is the letter of the element whose atomic number is closest to the element labeled "Z"?
- _____ 4) Which letter labels the halogen element with the lowest atomic number?
- _____ 5) Which of these elements is the least reactive? **A)** "J" **B)** "Y" **C)** "A" **D)** "O" **E)** "B"
- _____ 6) Which of these elements is/are NOT transition elements/metals. (list ALL that are not transition elements)? **A)** "C" **B)** "N" **C)** "K" **D)** "N" **E)** "Y"

continued on other side

The Periodic Table Worksheet Answer Key Section 2

Rex M. Heyworth, J. G. R. Briggs



The Periodic Table Worksheet Answer Key Section 2:

Glencoe Science McGraw-Hill Staff,2001-06 *Periodic Table* Cindy Blobaum,2005 What do chocolate chip cookies chemistry and logic have in common They are the basis for a unit that lets students become actively engaged in discovering the arrangement of the periodic table This learning activity takes the periodic table out of the static presentation usually associated with textbooks and chemistry courses and interjects an element of discovery The two activities in this unit provide students with information that they have to arrange in organized charts In the process of creating the arrangements students will be involved in problem solving and will gain an appreciation for the scientific process of exploration and verification This dynamic unit meets national science standards in seven teaching and content areas Bring the periodic table to life with this hands on minds on unit Book jacket *The School Science Review* ,2003 [Educart ICSE Class 10 One-shot Question Bank 2026 Chemistry \(strictly for 2025-26 boards\)](#) Sir Tarun Rupani,2025-07-12 Fast track your Chemistry revision with this exam ready resource This One shot Question Bank by Sir Tarun Rupani is designed to help ICSE Class 10 students revise the complete Chemistry syllabus quickly and thoroughly It simplifies theory boosts numerical accuracy and ensures strong exam practice all aligned with the 2025 26 ICSE syllabus Key Features Strictly Based on ICSE 2025 26 Curriculum Complete chapter coverage including Periodic Table Chemical Bonding Acid Base Organic Chemistry and more One shot Format Each chapter includes concise concept notes chemical equations reactions and key diagrams for quick recall Complete Coverage of Question Types Includes objective short long answers equation based numerical and reasoning questions Chapterwise PYQs Included Practice with previous years ICSE board questions to understand trends and improve retention Solved Answers in ICSE Format Clear well structured solutions using proper units chemical symbols and balanced equations Smart Revision Focus Special tips to avoid common mistakes in writing reactions balancing equations and attempting numericals Why Choose This Book This Chemistry One shot by Sir Tarun Rupani is built for smart preparation whether you re revising at the last minute or practising throughout the term It helps you approach each question with clarity confidence and the precision needed to score high in the 2026 ICSE board exam **2025-26 RRB JE CBT Stage-2 Practice Book** YCT Expert Team , 2025 26 RRB JE CBT Stage 2 Practice Book 272 495 E This book contains 51 sets of practice set **Sif Chemistry NI Pwb 2e** Rex M. Heyworth,2007 *Holt Chemistry* Ralph Thomas Myers,2004 [Sif Chemistry Ol Pwb 2e](#) ,2007 [Chemistry](#) Carson-Dellosa Publishing,2015-03-16 Chemistry for grades 9 to 12 is designed to aid in the review and practice of chemistry topics Chemistry covers topics such as metrics and measurements matter atomic structure bonds compounds chemical equations molarity and acids and bases The book includes realistic diagrams and engaging activities to support practice in all areas of chemistry The 100 Series science books span grades 5 to 12 The activities in each book reinforce essential science skill practice in the areas of life science physical science and earth science The books include engaging grade appropriate activities and clear thumbnail answer keys Each book has 128 pages and 100 pages or more of reproducible content to help

students review and reinforce essential skills in individual science topics The series will be aligned to current science standards

Chemistry Insights Rex M. Heyworth, J. G. R. Briggs, 2008

Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11

Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction

Atoms, Molecules & Elements: The Periodic Table Gr. 5-8 George Graybill, 2015-10-01 This is the chapter slice The Periodic Table from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Students will label each part of the atom learn what compounds are and explore the patterns in the periodic table of elements to find calcium Ca chlorine Cl and helium He through hands on activities These and more science concepts are presented in a way that makes them more accessible to students and easier to understand Written to grade and using simplified language and vocabulary and comprised of reading passages student activities crossword word search comprehension quiz and color mini posters our resource can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom's Taxonomy and STEM initiatives

Microscale Chemistry John Skinner, 1997 Developing microscale chemistry experiments using small quantities of chemicals and simple equipment has been a recent initiative in the UK Microscale chemistry experiments have several advantages over conventional experiments They use small quantities of chemicals and simple equipment which reduces costs The disposal of chemicals is easier due to the small quantities Safety hazards are often reduced and many experiments can be done quickly Using plastic apparatus means glassware breakages are minimised Practical work is possible outside a laboratory Microscale Chemistry is a book of such experiments designed for use in schools and colleges and the ideas behind the experiments in it come from many sources including chemistry teachers from all around the world Current trends indicate that with the likelihood of further environmental legislation the need for microscale chemistry teaching techniques and experiments is likely to grow This book should serve as a guide in this process

Chemistry, 2015-03-16 Chemistry for grades 9 to 12 is designed to aid in the review and practice of chemistry topics Chemistry covers topics such as metrics and measurements matter atomic structure bonds compounds chemical equations molarity and acids and bases The book includes realistic diagrams and engaging activities to support practice in all areas of chemistry The 100 Series science books span grades 5 to 12 The activities in each book reinforce essential science skill practice in the areas of life science physical science and earth science The books include engaging grade appropriate activities and clear thumbnail answer keys Each book has 128 pages and 100 pages or more of reproducible content to help students review and reinforce

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2023-24)Onward Dr. R. Goel, 2023-05-20 The series Learning Elementary Chemistry for Classes 6 to 8 has been revised strictly according to the latest curriculum The content of this series has been developed to fulfill the requirement of all the six domains Concepts Processes Applications Attitudes Creativity and World view of Science to make teaching and learning of Chemistry interesting understandable and enjoyable for young minds This series builds a solid foundation for young learners to prepare them for higher classes The main strength of the series lies in the subject matter and the experience that a learner will get in solving difficult and complex problems of Chemistry Emphasis has been laid upon mastering the fundamental principles of Chemistry rather than specific procedures Unique features of this series are The content of the

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Creating the Global Classroom Laurence Peters, 2022-03-10 The book examines how to begin to think like a global educator first by examining how our own histories and experiences have formed our own cultural and professional identities and second how the varied frames by which global education can be understood pedagogical ideological and cosmopolitan have shaped the field Laurence Peters connects theory and practice about global education relevant to cultivating global awareness in primary and secondary students Rather than seeing global education as a special field separate from the other disciplines the author encourages integration of global perspectives into everything we do Showcasing how global awareness is a developmental issue dependent upon the student s ability to step outside of their own place based comfort zone this volume lays out a roadmap of major challenges and issues around instilling this awareness in students This book connects theory and practice about global education relevant to cultivating global awareness in primary and secondary students From this foundation the book engages with the challenge of integrating global perspectives within a crowded curriculum By convincing students and teachers alike of global education s centrality thinking globally becomes an integral component of learning across subject areas and grade levels and this work encourages students to exercise empathy for the other and to develop critical skills to see through media distortions and fake news so they can better resist the tendency of politicians in our increasingly multicultural countries to divide people along racial and ethnic lines

Resources in Education ,1987 Serves as an index to Eric reports microform Holt Science and Technology 2002 Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2002

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