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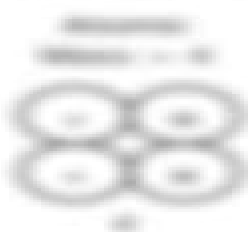
ATOMIC NUMBERS, PROTONS, NEUTRONS AND ELECTRONS

The number of protons in each atom of an element is its atomic number. The atomic number is also the number of electrons since atoms themselves are electronically neutral. Because neutrons and protons are almost the same mass, the total number of protons and neutrons in an atom is the atomic mass. Therefore, by subtracting the atomic number from the atomic mass you can calculate the number of neutrons.

This is why the mass number and atomic number, but not the number of neutrons, is shown in the periodic table. You can use the information provided below to determine the missing values in the chart below.



Example: Hydrogen atom



Use this knowledge and the information provided below to determine the missing values in the chart below.

Element	Symbol	Atomic Number	Atomic Mass	Protons	Neutrons	Electrons
Hydrogen	H					
Helium	He					
Lithium	Li					
Boron	B					
Carbon	C					
Nitrogen	N					
Oxygen	O					
Fluorine	F					
Neon	Ne					
Sodium	Na					
Magnesium	Mg					
Aluminum	Al					
Silicon	Si					
Phosphorus	P					
Sulfur	S					
Chlorine	Cl					
Argon	Ar					
Potassium	K					
Calcium	Ca					
Scandium	Sc					
Titanium	Ti					
Vanadium	V					
Chromium	Cr					
Manganese	Mn					
Iron	Fe					
Cobalt	Co					
Nickel	Ni					
Copper	Cu					
Zinc	Zn					
Gallium	Ga					
Germanium	Ge					
Arsenic	As					
Selenium	Se					
Bromine	Br					
Krypton	Kr					
Rubidium	Rb					
Strontium	Sr					
Yttrium	Y					
Zirconium	Zr					
Niobium	Nb					
Molybdenum	Mo					
Technetium	Tc					
Ruthenium	Ru					
Rhodium	Rh					
Palladium	Pd					
Silver	Ag					
Cadmium	Cd					
Indium	In					
Tin	Sn					
Antimony	Sb					
Tellurium	Te					
Iodine	I					
Xenon	Xe					
Cesium	Cs					
Barium	Ba					
Lanthanum	La					
Cerium	Ce					
Praseodymium	Pr					
Neodymium	Nd					
Europium	Eu					
Gadolinium	Gd					
Terbium	Tb					
Dysprosium	Dy					
Ho	Ho					
Er	Er					
Tm	Tm					
Yb	Yb					
Lu	Lu					
Hf	Hf					
Ta	Ta					
W	W					
Re	Re					
Os	Os					
Ir	Ir					
Pt	Pt					
Au	Au					
Hg	Hg					
Tl	Tl					
Pb	Pb					
Bi	Bi					
Po	Po					
At	At					
Rn	Rn					
Ac	Ac					
Th	Th					
Pa	Pa					
U	U					
Np	Np					
Pu	Pu					
Am	Am					
Cm	Cm					
Bk	Bk					
Cf	Cf					
Es	Es					
Fm	Fm					
Md	Md					
No	No					
Lr	Lr					

Where Are The Electrons Answer Key

SATYAM SIR



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From Electrons to Elephants and Elections Shyam Wuppuluri, Ian Stewart, 2022-04-08 This highly interdisciplinary book covering more than six fields from philosophy and sciences all the way up to the humanities and with contributions from eminent authors addresses the interplay between content and context reductionism and holism and their meeting point the notion of emergence Much of today's science is reductionist bottom up in other words behaviour on one level is explained by reducing it to components on a lower level Chemistry is reduced to atoms ecosystems are explained in terms of DNA and proteins etc This approach fails quickly since we cannot extrapolate to the properties of atoms solely from Schrödinger's equation nor figure out protein folding from an amino acid sequence or obtain the phenotype of an organism from its genotype An alternative approach to this is holism top down Consider an ecosystem or an organism as a whole seek patterns on the same scale Model a galaxy not as 400 billion point masses stars but as an object in its own right with its own properties spiral elliptic Or a hurricane as a structured form of moist air and water vapour Reductionism is largely about content whereas holistic models are more attuned to context Reductionism content and holism context are not opposing philosophies in fact they work best in tandem Join us on a journey to understand the multifaceted dialectic concerning this duo and how they shape the foundations of sciences and humanities our thoughts and the very nature of reality itself

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