

8 Structure and nomenclature of organic compounds

Unit	4	How are organic compounds categorised, analysed and used?
Area of Study	1	How can the diversity of carbon compounds be explained and categorised?
Topic	8	Structure and nomenclature of organic compounds
Subtopic	8.2	The carbon atom

Summary

- Carbon atoms have 4 valence electrons and make 4 covalent bonds with other carbon atoms or atoms of different elements.
- Bond energy is the amount of energy required to break the bonds of reactants or the amount of energy released per mole when a covalent bond forms between 2 molecules. It is typically measure in kJ mol^{-1} .
- The bond length is the distance between the nuclei of the atoms sharing the electrons.
- Carbon molecules with single bonds typically form a tetrahedral arrangement.
- Double $\text{C}=\text{C}$ bonds are stronger than $\text{C}-\text{C}$ single bonds and forms flat molecules. $\text{C}\equiv\text{C}$ bonds are stronger again and form linear molecules.



My notes

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**Robert Stokes,Angela Stubbs,Neale
Taylor,Jason Bourke,Ben
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Chandler, Michele Herrington, Lam Pham, 2022-02-25 This book establishes a new theoretical and practical framework for multimodal disciplinary literacy MDL fused with the subject specific science pedagogies of senior high school biology chemistry and physics It builds a compatible alignment of multiple representation and representation construction approaches to science pedagogy with the social semiotic systemic functional linguistic based approaches to explicit teaching of disciplinary literacy The early part of the book explicates the transdisciplinary negotiated theoretical underpinning of the MDL framework followed by the research informed repertoire of learning experiences that are then articulated into a comprehensive framework of options for the planning of classroom work Practical adoption and adaptation of the framework in biology chemistry and physics classrooms are detailed in separate chapters The latter chapters indicate the impact of the collaborative research on teachers professional learning and students multimodal disciplinary literacy engagement concluding with proposals for accommodating emerging developments in MDL in an ever changing digital communication world The MDL framework is designed to enable teachers to develop all students disciplinary literacy competencies This book will be of interest to researchers teacher educators and postgraduate students in the field of science education It will also have appeal to those in literacy education and social semiotics The Open Access version of this book available at www.taylorfrancis.com has been made available under a Creative Commons Attribution Non Commercial No Derivatives 4.0 license **VCE Chemistry Study Guide to Unit 2**, 1992 *A+ Chem Notes VCE Unit 3* Jennifer Willis, 2012 The third edition of Chem Notes VCE Unit 3 has been written to precisely match the VCE Chemistry Study Design to be implemented in 2013 This edition includes a data book and is 100% up to date Chem Notes VCE Unit 3 includes a Comprehensive summary notes on chemical analysis and organic chemical pathways a Practice exam questions a Solutions a Data sheets a

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