

Energy and Thermal Energy Review

Thermal Energy	1. the sum of the kinetic and potential energy of the molecules of a substance
Conduction	2. transfer of heat by particles colliding with each other
Convection	3. transfer of heat by the movement of a gas or a liquid
Radiation	4. transfer of energy by electromagnetic waves
Thermometer	5. an instrument used to measure temperature
Conductor	6. a substance that conducts heat
Insulator	7. a material through which heat does not easily flow
Heat	8. thermal energy transferred because of a difference in temperature
Fahrenheit	9. temperature scale with 180 degrees between the freezing and boiling of H ₂ O
Celsius	10. temperature scale with 100 degrees between the freezing and boiling of H ₂ O
Temperature	11. the average kinetic energy of the particles of a substance
Specific Heat	12. amount of heat to raise 1 kg of a substance 1°C

13. As matter is heated, its particles move **faster**.

14. Particles that make up matter are in a state of **constant motion**.

15. The energy of moving particles is called **kinetic energy**.

16. The energy stored inside a battery is **chemical** energy.

17. Temperature is a measure of the **average kinetic energy** of atoms and molecules.

18. When a liquid is heated it will (expand or contract)? **expand**

19. A pot on a stove gets hot by means of **conduction**.

20. Since water has a higher specific heat than sand, water will heat **slower** than sand.

21. What are the energy transformations in the follow: a nuclear reaction generates heat to produce steam. The steam is used to turn a generator which finally produces electricity.

Nuclear to Thermal to Kinetic to Energy

22. Photosynthesis transforms electromagnetic energy from the Sun into **chemical** energy.

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Mike Jess



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