

## CHAPTER - 04

# CHEMICAL BONDING AND MOLECULAR STRUCTURE

### Chemical bond

It is the force of attraction between the constituent particles like ions, atoms, metal ions and molecules, in order to get the stability.

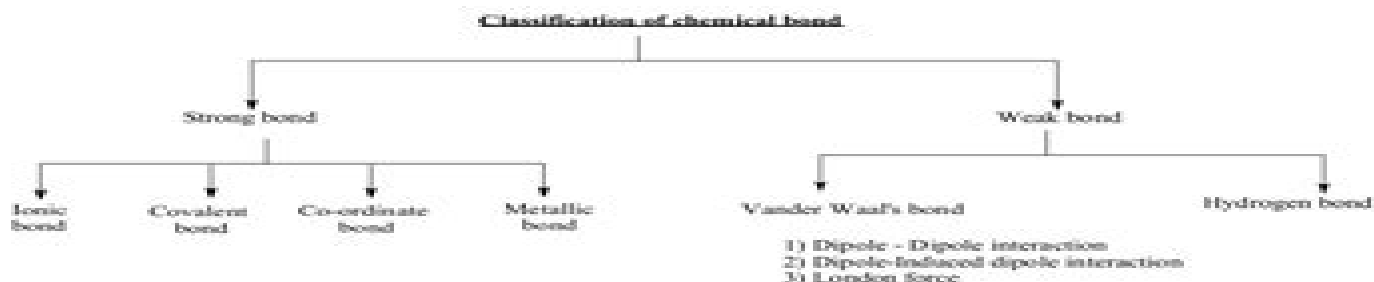
- Chemical bonding process accompanied by decrease in energy.
- Decrease in energy  $\propto$  strength of the bond.

### Causes of chemical combination

#### I. Tendency to acquire minimum energy.

1. When two atoms approaches to each other. There is attractive forces between nucleus and electron and vice versa.
2. Also repulsive forces present between nucleus and electrons of two atoms.
3. Attractive forces are dominant than repulsive force, therefore total energy of the system decreases.
4. Stability  $\propto \frac{1}{\text{energy}}$   $\propto$  attractive force  $\propto \frac{1}{\text{repulsive force}}$
5. Bond formation is exothermic.

#### II. Tendency to acquire stable configuration



# Study Island Chemical Bonding

**S Baum**



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