Polyatomic Ions, Ionic Reactions and Redox Reactions

- Polyatomic ions are covalent molecules that possess charge and behave as normal ions in solution.
- When a salt containing a polyatomic ion is dissolved in water, the polyatomic ions themselves <u>DO NOT</u> break apart. They are simply separated from the *counter-ion*.
- Example: Phosphate (PO₄3-)

$$Na_3PO_4$$
 (s) $\xrightarrow{H_2O(L)}$ \Rightarrow $3Na^*$ (aq) + PO_4^{3-} (aq)

Sodium Phosphate cations Phosphate

What Is A Polyatomic Ion Packets Chemical Reactions

Morris Hein, Susan Arena, Cary Willard

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