PHYSICS GRADE 10

CHAPTER ONE LINEAR MOTION

Anthordust Nove

toucky of mostlers is all violed linear owns;

- 1. Character Side.
- 2. Oyunamina

in kinematics forces causing motion are disregarded while dynamics deals with motion of objects and the forces sausing them.

Classification of the control of the c

Distance moved by a body is a specified direction is called displacement, it is denoted by letter's' and has both magnitude and direction. Distance is the ineventent from one point to another. The Si unit for all valuement is the matter test.

St. Samed

This is the distance supered are well time.

Speeds distance covered/ time taken. Obtains is a scalar quantity since it has magnitude only. The Si unit for speed is metres are second lim/s or ess't

American appends total distance covered/total time taken

stationer would a free schemed uppent one stimule.

of the second second

 A budy covers a distance of 20m in A seconds. A rests for 10 seconds and finally covers a distance of 50m in 60 seconds. Calculate the ownesse speed.

North Service

flast of stance never educated at down

The all fillers is a knowled with a match supported.

The effect everage speeds 100/20-5 m/s.

 Colculate the distance in metres covered by a body moving with a uniform speed of 140 km/h in 30 seconds.

Name of Street

Other proper oder promise appropriate town.

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+18871889/89769-038//s
-69738
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14 C / A SECURE

 Calculate the time in seconds taken a by body moving with a uniform speed of 160km/h to cover a distance of 1,000 km?

Secretary of the last of the l

Reprint the kind has been been been been a point of

Mirror willful amount species.

NAMES OF A STREET, STR

-00,000 seconds.

Mile Management

This is the change of displacement per unit time. It is a vector quantity,

Velocity-change in displacement/total time taken.

The 52 units for respectly are make

All professional dispersion

What To Study For Grade 10 Physics Nov 2014 P1

Barbara Bridgman Perkins

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Cancer, Radiation Therapy, and the Market Barbara Bridgman Perkins, 2017-08-16 Appraising cancer as a major medical market in the 2010s Wall Street investors placed their bets on single technology treatment facilities costing 100 300 million each Critics inside medicine called the widely publicized proton center boom crazy medicine and unsustainable public policy. There was no valid evidence they claimed that proton beams were more effective than less costly alternatives. But developers expected insurance to cover their centers staggeringly high costs and debts. Was speculation like this new to health care. Cancer Radiation Therapy and the Market shows how the radiation therapy specialty in the United States later called radiation oncology coevolved with its device industry throughout the twentieth century Academic engineers and physicians acquired financing to develop increasingly powerful radiation devices initiated companies to manufacture the devices competitively and designed hospital and freestanding procedure units to utilize them In the process they incorporated market strategies into medical organization and practice Although palliative benefits and striking tumor reductions fueled hopes of curing cancer scientific research all too often found serious patient harm and disappointing beneficial impact on cancer survival This thoroughly documented and provocative inquiry concludes that public health policy needs to re evaluate market driven high tech medicine and build evidence based health care systems

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the boiling point of an organic liquid is the temperature at which its vapour pressure equals the atmospheric pressure over the liquid or it is the temperature at which the vapour and liquid phases are in equilibrium at agiven pressure the boiling point is considered as a criterion of purity of a compound and is useful for identification of

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given that the boiling points of ethanol and methanol 78 c and 64 c respectively are far off in a significant way this seems like a valid conclusion conclusion thermometer calibrations are necessary in order to collect reliable information as precise calculations are imperative to gathering correct results

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melting point 1 a push the open end of a capillary tube into the powdered acetamide b move the powder to the closed end of the capillary tube by tapping it on the table repeat until the the powdered acetamide occupies 1 2 mm of the capillary tube end c

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discussion and conclusion the technique recrystallization was used because it is a simple method to purify a solid compound the percentage yield for benzoic acid and acetanilide acid being high indicated the initial sample had a low concentration of impurities with the melting point being in a specific

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introduction in this lab we will be using boiling points melting points and infrared spectrometry to identify unknown compounds boiling points are affect by three key factors the first being intermolecular forces inter molecular affect boiling point based on bond strength

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determination of boiling points introduction the boiling point of a compound is the temperature at which it changes from a liquid to a gas this is a physical property often used to identify substances or to check the purity of the compound it **boiling point determination lab report introduction the** - Jul 19 2023

clean up solid and liquid spills immediately true g conclusions after conducting the boiling point determination lab it can be concluded that the unknown substance 193 is isopropyl alcohol on account of its observed boiling point of 88 c

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conclusion after completing this experiment the melting point of three different unknown compounds was obtained using the mel temp apparatus compound a had a melting point range of 114 6 115 5 c thus revealing it to be acetanilide compound b had a melting point range of 122 6 124 5 c thus

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nov 27 2019 essay sample the purpose of this experiment is to determine the boiling points of various organic compounds and to use these to identify unknowns equipment free essays

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melting point can be used as the identity and purity of an organic solid however it is not the same as the boiling point the boiling point is a characteristic unique of pure liquids the boiling point is the temperature at which the vapor pressure of a liquid is equal to the atmospheric pressure

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the purpose of the experiment is to understand what vapour pressure is and how it changes with temperature to understand

the relationship between vapour pressure at a given temperature and boiling temperature at a given pressure to determine the vapour pressure of a pure liquid at various temperatures

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exploring boiling points abstract in this experiment we are given a set of straight chain alkanes and a set of compounds tocompare by following the trendline of the straight chain alkanes it could be concluded that as the molar mass of the straight chain alkanes increases their boiling points increase as well

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laboratory partners samantha and mackenzie goals \square correctly determine the boiling point of an unknown liquid \square measure the refractive index of an unknown \square use the gas chromatogram to determine the mass percentage of a mixture and boiling point determination faculty of science - Jun 18 2023

50 c so in order to convert an experimental measurement taken in calgary which is at higher altitude than sea level and so is at a lower pressure to that reported for sea level higher pressure one needs to add a correction factor since the boiling point at sea level is higher than that at higher altitudes

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introduction in chemistry one may encounter an unknown substance that need to be determined by physical properties such as boiling point melting point density solubility etc in this experiment i am to figure out the boiling point of a mystery substance in order to identify it chemical responsibility this experiment requires the handling of a bunsen burner boiling 6 2b step by step procedures for boiling point determination - Mar 15 2023

there are a variety of methods by which a sample s boiling point can be determined including distillation reflux and by using a thiele tube the most straightforward method uses a thiele tube and has the advantage of using less than 0 5ml 0 5 ml of material

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