



Trends in Atomic Radii

1. Describe and explain the trend in atomic radii as you go down group 2.
2. Describe and explain the trend in atomic radii as you go across period 3.
3. Describe and explain the trend in atomic radii as you go across period 4 from Scandium to Zinc.

Radii Comparisons

4. Compare the radii of the following pairs, including both a description and an explanation in your answer.
 - a. Na and Na^+
 - b. Li and O^-
 - c. Al^{3+} and Na^+
 - d. H^+ and H^-
 - e. S^{2-} and Cl^-
 - f. F and Na^+



Teaching Transparency 2atomic And Ionic Radii

Holt Rinehart & Winston



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