

વિભાગ A

[50]

- નીચે આપેલા પાંચ જવાબો પૈકી સાચો વિકલ્પ પસંદ કરો. (પ્રત્યેકના 1 મુલ્ય)
1. ZnO ના સ્ફટિકને તરમ કરતા તેનું સૂત્ર અને છે.
(A) $Zn_{1-x}O$ (B) $Zn_{1+x}O$ (C) ZnO (D) આપેલ તમામ
 2. અંતઃકેન્દ્રિત એકમકોષ રચનાની પેકિંગ ક્ષમતા હોય છે.
(A) 40% (B) 68% (C) 72% (D) 74%
 3. Fe_xO ના $Fe(II)$ માટે એક $Fe(III)$ પડાવે છે, x નું મૂલ્ય શું હશે ?
(A) $\frac{8}{9}$ (B) $\frac{2}{3}$ (C) $\frac{3}{4}$ (D) $\frac{5}{3}$
 4. સામાન્ય તાપમાને સોડિયમ પાતુ અંતઃકેન્દ્રિત પત્રમાં સ્ફટિકીકરણ પામે છે. એકમ કોષની પાર $a = 4.29 \text{ \AA}$ છે. સોડિયમ પરમાણુની નિજ્યા છે.
(A) $1.40 \text{ \AA}$ (B) $2.65 \text{ \AA}$ (C) $1.85 \text{ \AA}$ (D) $2.15 \text{ \AA}$
 5. પોટાશ એલમનું અણુસૂત્ર
(A) $K_2SO_4 \cdot Al(SO_4)_3 \cdot 24 H_2O$ (B) $K_2SO_4 \cdot Al_2(SO_4)_3 \cdot 24 H_2O$
(C) $K_2SO_4 \cdot Al_2(SO_4)_2 \cdot 24 H_2O$ (D) $K_2SO_4 \cdot Al_2(SO_4)_3 \cdot 24 H_2O$
 6. આબ્મસીલ પટક A અને B ના જલીય દ્રાવણનું સંતુલન સ્થિતિએ કુલ દબાણ 0.02 બાર છે. જો A પટકના મોલઅંશ 0.2 હોય, તો પટક B નું આંશિક દબાણ કેટલું હશે ?
(A) 0.02 બાર (B) 0.04 બાર (C) 0.16 બાર (D) 0.2 બાર
 7. 298 K તાપમાને CO_2 ના તેના સંતૃપ્ત જલીય દ્રાવણમાં મોલઅંશ કેટલા હશે ? [$K_H = 6.02 \times 10^{-4} \text{ bar}$, $P_{CO_2} = 2 \times 10^{-6} \text{ bar}$]
(A) 3×10^{-4} (B) 3.32×10^{-5} (C) 1.23×10^{-4} (D) 2.82×10^{-5}
 8. $MgSO_4$, $AgNO_3$ અને $AlCl_3$ ના દ્રાવણમાંથી 1 મોલ ઈલેક્ટ્રોન પસાર કરતાં કુલો આગળ Mg, Ag અને Al નીચેના પૈકી કયા મોલ મુક્તીનારમાં જમા થશે ?
(A) 1 : 1 : 1 (B) 1 : 2 : 3 (C) 2 : 1 : 3 (D) 3 : 6 : 2
 9. સાંદ $NaCl$ ના જલીય દ્રાવણનું ફ્રેઝરન્ટના કુલો વચ્ચે વિદ્યુતવિભાજન કરતાં કેથોડ અને એનોડ ઉપર અનુક્રમે શું ઉત્પન્ન થાય છે ?
(A) Cl_2 અને Na (B) H_2 અને O_2 (C) Cl_2 અને H_2 (D) H_2 અને Cl_2
 10. લેડ સંચાલક કોષમાં કયું દ્રાવણ (વિદ્યુતવિભાજન) ભરવામાં આવે છે ?
(A) 30%(w/w) સાંદ H_2SO_4 (B) 48%(w/w) સાંદ H_2SO_4
(C) 38%(w/w) સાંદ H_2SO_4 (D) 38%(w/w) સાંદ HNO_3
 11. હાર્ડમેન બળતણ કોષમાં એનોડ પર થતી પ્રક્રિયા કઈ છે ?
(A) $2H_{2(g)} + 4OH^-_{(aq)} \rightarrow 4H_2O_{(l)} + 4e^-$ (B) $O_{2(g)} + 2H_2O_{(l)} + 4e^- \rightarrow 4OH^-_{(aq)}$
(C) $Pb^{2+} + H_2O_{(l)} + \frac{1}{2} O_{2(g)} \rightarrow Pb(OH)_2 + 2e^-$ (D) $2H_{2(g)} + O_{2(g)} \rightarrow 2H_2O_{(l)} + 571.7 \text{ KJ}$
 12. એક અજ્ઞાત પ્રક્રિયામાં 2 મોલ પ્રક્રિયકના મોલમાંથી 1 મોલ નીપજ બનવામાં 1 કલાક લાગે છે. 4 કલાક પછીથી પ્રક્રિયકના કેટલા મોલ બાકી રહેશે ?
(A) 0.125 મોલ (B) શૂન્ય મોલ (C) 0.225 મોલ (D) 0.5 મોલ
 13. એક આજીવીય પ્રક્રિયાને ઊંચા દબાણે અથવા ઊંચા તાપમાને કરવાથી તેનો પ્રક્રિયા ક્રમ કેટલો થાય ?
(A) 0 (B) 2 (C) 1 (D) $\frac{3}{2}$
 14. સામાન્ય રીતે રાસાયણિક પ્રક્રિયાનો વેગ બમણો કરવા માટે તાપમાનમાં 10° નો વધારો કરાય છે. જો પ્રક્રિયાના તાપમાનમાં

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