

**CHAPTER 11 REVIEW***Molecular Composition of Gases***MIXED REVIEW**

**SHORT ANSWER** Answer the following questions in the space provided.

1. \_\_\_\_\_ The average speed of a gas molecule is most directly related to the \_\_\_\_\_.  
(a) polarity of the molecule  
(b) pressure of the gas  
(c) temperature of the gas  
(d) number of moles in the sample
2. \_\_\_\_\_ Which of the following statements best represents Graham's law for two gases at the same temperature?  
(a) The molar masses and the velocities are inversely proportional.  
(b) The molar masses and the velocities are directly proportional.  
(c) The molar masses and the square roots of the velocities are directly proportional.  
(d) The molar masses and the squares of the velocities are inversely proportional.
3. Water is synthesized from its elements according to the following equation:  
$$2\text{H}_2(g) + \text{O}_2(g) \rightarrow 2\text{H}_2\text{O}(l)$$
  - \_\_\_\_\_ a. Which is greater, the volume of  $\text{H}_2$  consumed or the volume of  $\text{O}_2$  consumed?
  - \_\_\_\_\_ b. Which is greater, the mass of  $\text{H}_2$  consumed or the mass of  $\text{O}_2$  consumed?
4. Consider a 2.24 L sample of an ideal gas at STP.
  - \_\_\_\_\_ a. How many moles are present in the sample, or is more data needed?
  - \_\_\_\_\_ b. How many molecules are present in the sample, or is more data needed?
  - \_\_\_\_\_ c. What is the mass of the sample, or is more data needed?

**PROBLEMS** Write the answer on the line to the left. Show all your work in the space provided.

5. \_\_\_\_\_ Calculate the volume of 3.5 mol of an ideal gas at 24°C and 760 torr.

# Skills Concept Review Molecular Composition Of Gases

**RJ Shavelson**



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Association, 2004 Fire fighter stories of dreams realized bravery tested and lives saved Twelve men and women who have devoted their lives to saving others tell their stories

**Concepts of Matter in Science Education** Georgios

Tsaparlis, Hannah Sevian, 2013-07-09 Bringing together a wide collection of ideas reviews analyses and new research on particulate and structural concepts of matter Concepts of Matter in Science Education informs practice from pre school through graduate school learning and teaching and aims to inspire progress in science education The expert contributors offer a range of reviews and critical analyses of related literature and in depth analysis of specific issues as well as new research Among the themes covered are learning progressions for teaching a particle model of matter the mental models of both students and teachers of the particulate nature of matter educational technology chemical reactions and chemical phenomena chemical structure and bonding quantum chemistry and the history and philosophy of science relating to the particulate nature of matter The book will benefit a wide audience including classroom practitioners and student teachers at every educational level teacher educators and researchers in science education If gaining the precise meaning in particulate terms of what is solid what is liquid and that air is a gas were that simple we would not be confronted with another book which while suggesting new approaches to teaching these topics confirms they are still very difficult for students to learn Peter Fensham Emeritus Professor Monash University Adjunct Professor QUT from the foreword to this book Scientific and Technical Aerospace Reports, 1970

**Elementary Principles of Chemical Processes** Richard M. Felder, Ronald W.

Rousseau, Lisa G. Bullard, 2020-08-11 This best selling text prepares students to formulate and solve material and energy balances in chemical process systems and lays the foundation for subsequent courses in chemical engineering The text provides a realistic informative and positive introduction to the practice of chemical engineering

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