In your textbook, read about water and its importance.

#### For each statement below, write true or false.

- In a water molecule, electrons are shared equally between the hydrogen atoms and oxygen atom.
  - The attraction of opposite charges between hydrogen and oxygen forms a weak oxygen bond.
- Because of its polarity, water can move from the roots of a plant up to its leaves.
  - 4. Water changes temperature easily.
    - 5. Unlike most substances, water expands when it freezes.

#### Circle the letter of the choice that best completes the statement.

- 6. All objects in motion have
  - a. potential energy.
     b. heat energy.
- c. kinetic energy.

d. diffusion.

- d. random energy.
- 7. The first scientist to observe evidence of the random motion of molecules was
  - a. Brown.
- b. Darwin.

- c. Mendel. d. Hooke.
- The net movement of particles from an area of higher concentration to an area of lower concentration is called
  - a. dynamic equilibrium.
  - c. concentration gradient.
- 9. Diffusion occurs because of
  - a. nonrandom movement of particles.
  - a chemical reaction between particles.
- b. random movement of particles.

b. nonrandom movement.

- chemical energy.
- 10. When a few drops of colored corn syrup are added to a beaker of pure corn syrup, the color will
  - a. move from low concentration to high concentration.
  - b. form a polar bond.
  - c. start to diffuse.
  - d. remain on the bottom of the beaker.
- 11. Diffusion can be accelerated by
  - a. decreasing the pressure.

- **b.** increasing the temperature.
- decreasing the movement of particles.
- increasing the dynamic equilibrium.
- 12. When materials pass into and out of a cell at equal rates, there is no net change in concentration inside the cell. The cell is in a state of
  - a. dynamic equilibrium.
     b. metabolism.
- imbalance.

- d. inertia.
- 13. The difference in concentration of a substance across space is called
  - a. dynamic equilibrium.

b. concentration gradient.

c. diffusion.

d. Brownian movement.

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**Prentice-Hall Staff** 

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