

Skills Worksheet

Concept Review Chapter 5: Section 1

Section: Simple Ions

Complete each statement below by choosing a term from the following list. Terms may be used more than once.

10

11

ion

octet

12

13

anion

cation

1. An atom or group of atoms that has a positive or negative electric charge because it has lost or gained electrons is a(n) ion.
2. The sodium ion has 11 protons, 12 neutrons, and 10 electrons.
3. In most chemical reactions, atoms tend to match the outer electron configuration of the noble gases. This is called the octet rule.

Complete each statement below by writing the correct term or phrase.

4. An anion is an ion with a negative charge.
5. Valence electrons occupy the outermost energy level of an atom.
6. The chloride ion and the chlorine atom have different electron configurations.
7. A cation is an ion with a positive charge.
8. Many stable ions have an electron configuration of a noble gas.

Answer the following questions in the space provided.

9. How do the outer-shell electron configurations for ions of Group 1, Group 2, and Group 15, Group 16, and Group 17 elements compare with those of the noble gases?

They are the same. Group 1 + 2 ions have noble gas configuration of previous row. Group 15, 16 + 17 ions have noble gas configuration in their row.

World Of Chemistry Section Review 8 3 Answer Key

Michael B. McElroy



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Table of Contents World Of Chemistry Section Review 8 3 Answer Key

1. Understanding the eBook World Of Chemistry Section Review 8 3 Answer Key
 - The Rise of Digital Reading World Of Chemistry Section Review 8 3 Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying World Of Chemistry Section Review 8 3 Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an World Of Chemistry Section Review 8 3 Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from World Of Chemistry Section Review 8 3 Answer Key
 - Personalized Recommendations
 - World Of Chemistry Section Review 8 3 Answer Key User Reviews and Ratings
 - World Of Chemistry Section Review 8 3 Answer Key and Bestseller Lists
5. Accessing World Of Chemistry Section Review 8 3 Answer Key Free and Paid eBooks
 - World Of Chemistry Section Review 8 3 Answer Key Public Domain eBooks
 - World Of Chemistry Section Review 8 3 Answer Key eBook Subscription Services
 - World Of Chemistry Section Review 8 3 Answer Key Budget-Friendly Options
6. Navigating World Of Chemistry Section Review 8 3 Answer Key eBook Formats

- ePub, PDF, MOBI, and More
 - World Of Chemistry Section Review 8 3 Answer Key Compatibility with Devices
 - World Of Chemistry Section Review 8 3 Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of World Of Chemistry Section Review 8 3 Answer Key
 - Highlighting and Note-Taking World Of Chemistry Section Review 8 3 Answer Key
 - Interactive Elements World Of Chemistry Section Review 8 3 Answer Key
8. Staying Engaged with World Of Chemistry Section Review 8 3 Answer Key
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers World Of Chemistry Section Review 8 3 Answer Key
9. Balancing eBooks and Physical Books World Of Chemistry Section Review 8 3 Answer Key
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection World Of Chemistry Section Review 8 3 Answer Key
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine World Of Chemistry Section Review 8 3 Answer Key
- Setting Reading Goals World Of Chemistry Section Review 8 3 Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of World Of Chemistry Section Review 8 3 Answer Key
- Fact-Checking eBook Content of World Of Chemistry Section Review 8 3 Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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